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Search for a space charge layer in thin film battery materials with low-energy muons

Jun Sugiyama^{1,*}, Elisabetta Nocerino², Ola K. Forslund³,
Yasmine Sassa³, Martin Månsson², Shigeru Kobayashi⁴,
Kazunori Nishio⁴, Taro Hitosugi⁴, Andreas Suter⁵, Thomas Prokscha⁵

¹ Neutron Science and Technology Center, Comprehensive Research Organization for Science and Society (CROSS), Tokai, Ibaraki 319-1106, Japan

² Department of Applied Physics, KTH Royal Institute of Technology, Roslagstullsbacken 21, SE-106 91 Stockholm, Sweden

³ Department of Physics, Chalmers University of Technology, SE-412 96 Göteborg, Sweden

⁴ School of Materials and Chemical Technology, Tokyo Institute of Technology, Tokyo 152-8552, Japan

⁵ Laboratory for Muon Spin Spectroscopy, Paul Scherrer Institut, CH-5232 Villigen PSI, Switzerland

E-mail: juns@triumf.ca or j.sugiyama@cross.or.jp

Abstract. In an all solid state Li-ion battery, it is crucial to reduce ionic resistivity at the interface between the electrode and the electrolyte in order to enhance Li^+ mobility across the interface. Recent first principles calculations predict the presence of a space-charge layer (SCL) at the interface due to the difference in the Li^+ chemical potential at the interface between two different materials, as in the metal-semiconductor junction in electronic devices. However, the presence of SCL has never been experimentally observed. Our first attempt in a fresh multilayer sample, $\text{Cu}(10\text{ nm})/\text{Li}_3\text{PO}_4(50\text{ nm})/\text{LiCoO}_2(100\text{ nm})$ on a sapphire substrate, with low-energy $\mu^+\text{SR}$ (LE $\mu^+\text{SR}$) revealed a gradual change in the nuclear magnetic field distribution width as a function of implantation depth even across the interface between Li_3PO_4 and LiCoO_2 . This implies that the change in the field distribution width at SCL of the sample is too small to be detected by LE $\mu^+\text{SR}$.

1. Introduction

The all solid-state battery, consisting of a solid cathode, solid anode, and solid electrolyte, is believed to provide a solution for several issues concerning current Li-ion batteries, as it would allow to improve safety and volumetric charge density of the current Li-ion batteries. For this reason, in recent years massive attention has been directed towards the development of all solid-state batteries from the scientific community. The most significant issue to overcome for realizing an all solid-state battery is how to control the interface layer, which is newly formed during a charge and discharge reaction, between the electrode and electrolyte. In fact, according to electrochemical analyses and ex-situ compositional analyses, such a layer, i.e., a solid-electrolyte interface (SEI), is known to exist in conventional Li-ion batteries based on liquid electrolyte [1]. A similar layer is expected to be also formed in an all solid-state battery [2].

Very recent first principles calculations highlighted the possible occurrence of a second issue [3]: the formation of a space-charge layer (SCL) at the interface between the cathode and



electrolyte due to the difference of their chemical potentials. In other words, Li vacancies are spontaneously formed in the electrolyte layer at the vicinity of the interface. Furthermore, the vacancy concentration is expected to decrease with the distance from the interface and finally become negligibly small. Such an SCL naturally increases the interfacial resistance, leading to the decrease in both the working voltage and response time for a rapid charge/discharge. In the worst case scenario, an all solid-state battery would not work at all due to the SCL.

At present, the existence of such layer has not been observed experimentally. However, empirical indications of the existence of SCL are provided, for example, by the LiCoO_2 battery material. Indeed, an oxide buffer layer interposed between the LiCoO_2 cathode and the sulfide electrolyte significantly reduces the interfacial resistance [4]. In addition, electric potential distribution measurements in an all solid-state battery ($\text{LiCoO}_2/\text{Li}_{1+x+y}\text{Al}_y\text{Ti}_{2y}\text{Si}_x\text{P}_{3x}\text{O}_{12}/\text{Li}$) showed a potential gradient in the electrolyte layer within approximately 1000 nm of the interface [5]. Nevertheless, more reliable or direct evidence of the SCL is required to further improve the interface properties in an all solid-state battery.

In order to observe the SCL, a nondestructive technique with a good depth resolution is needed, since the attempt to fabricate the samples for cross-section-view observations alters and/or destroys the potential gradient at the interface. While a conventional μ^+ SR is one of the powerful tools to observe the Li-distribution and diffusion in solids [6, 7], it lacks a depth resolution. We have, therefore, attempted to use a low-energy (LE) μ^+ SR technique, which allows depth dependent studies with an adjustable range, by tuning the implantation energy (E_{imp}) down to keV range [8, 9, 10, 11]. In the SCL, the relaxation rate of the transverse field (TF) oscillation spectrum is expected to decrease, due to the decreasing Li concentration. Here, TF means the magnetic field perpendicular to the initial muon spin polarization. Thus, we have mainly measured the TF- μ^+ SR spectra as a function of E_{imp} .

2. Experimental

The following two multilayers samples were prepared by a PLD technique:

#1: $\text{Cu}(10 \text{ nm})/\text{Li}_3\text{PO}_4(50 \text{ nm})/\text{LiCoO}_2(100 \text{ nm})$ on a sapphire substrate

#2: $\text{Cu}(10 \text{ nm})/\text{Li}_3\text{PO}_4(75 \text{ nm})/\text{LiCoO}_2(100 \text{ nm})$ on a sapphire substrate

The area of the sample was $10 \times 10 \text{ mm}^2$. The multilayers sample was placed on the Ni coated Al plate. The LE μ^+ SR spectra were measured mainly in TF with $H_{\text{TF}} = 50 \text{ Oe}$ at temperatures between 100 and 320 K. The experimental techniques are described in more detail elsewhere [12, 13]. The obtained μ^+ SR spectra were analyzed using *musfit* [14].

3. Results and Discussion

Figures 1(a) and 1(b) show the volume fraction (V_F) of the stopped muons as a function of E_{imp} in the sample #1 simulated with SRIM [15] and TRIMSP [16], while Fig. 1(c) shows the E_{imp} dependence of the TF relaxation rate (σ_{TF}) at 320 K. Here, σ_{TF} was obtained by fitting the TF- μ^+ SR spectra with:

$$A_0 P_{\text{TF}}(t) = A_{\text{TF}} \cos(2\pi f_{\text{TF}} t + \phi_{\text{TF}}) \exp(-\frac{1}{2} \sigma_{\text{TF}}^2 t^2), \quad (1)$$

where A_0 denotes the initial asymmetry at $t = 0$, $P_{\text{TF}}(t)$ denotes the μ^+ spin depolarization function, $A_{\text{TF}} (= A_0)$ denotes the asymmetry, f_{TF} denotes the μ^+ spin precession frequency caused by TF, ϕ_{TF} denotes the initial phase of the precession, and σ_{TF} denotes the Gaussian relaxation rate for the oscillatory signal and roughly corresponds to the spin-spin relaxation rate.

From the $V_F(E_{\text{imp}})$ curves simulated with SRIM [Fig. 1(a)], it is found that $V_F^{\text{Cu}} \sim 100\%$ at $E_{\text{imp}} = 2 \text{ keV}$, $V_F^{\text{Li}_3\text{PO}_4} \sim 100\%$ at $E_{\text{imp}} = 8 \text{ keV}$, and $V_F^{\text{LiCoO}_2} \sim 100\%$ at $E_{\text{imp}} = 20 \text{ keV}$.

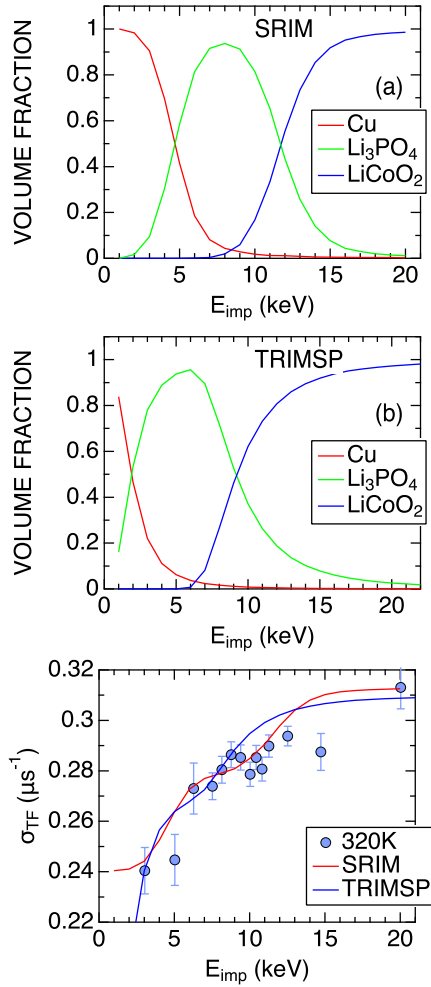


Figure 1. The volume fraction of the implanted μ^+ stopped in the three layers in the sample #1, Cu(10 nm)/Li₃PO₄(50 nm)/LiCoO₂(100 nm) on a sapphire substrate, as a function of the implantation energy (E_{imp}) simulated with (a) SRIM [15] and (b) TRIMSP [16]. (c) the relationship between the TF relaxation rate (σ_{TF}) and E_{imp} for the sample #1. In (c), the predicted $\sigma_{\text{TF}}(E_{\text{imp}})$ curves with SRIM and TRIMSP are also plotted.

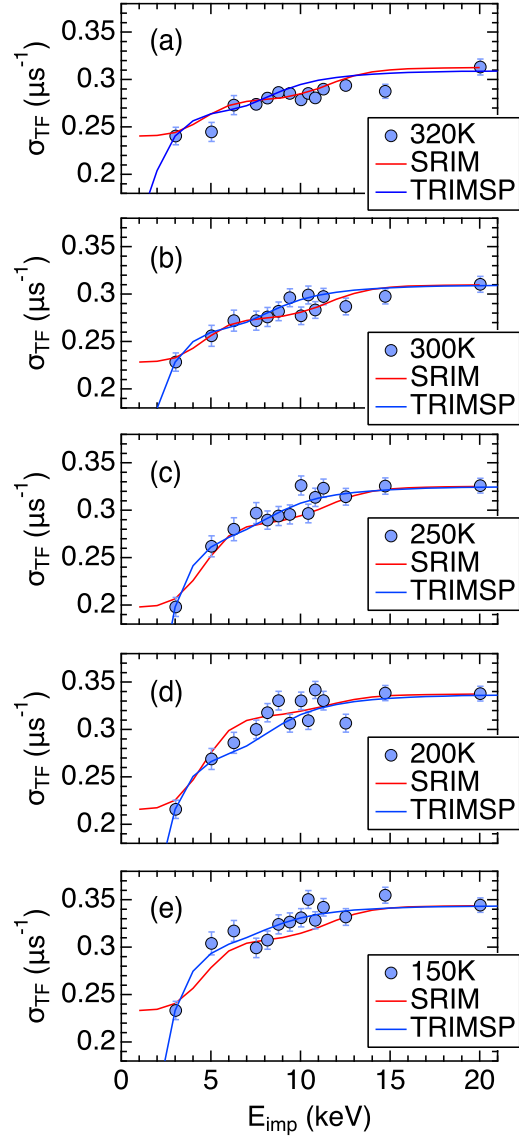


Figure 2. The E_{imp} dependence of σ_{TF} for the sample #1 measured at (a) 320 K, (b) 300 K, (c) 250 K, (d) 200 K, and (e) 150 K. Red solid lines represent the predicted dependence with SRIM, while blue solid line represents that with TRIMSP only in (a). (a) is the same to Fig. 1(c).

Therefore, using the $V_{\text{F}}(E_{\text{imp}})$ curves predicted with SRIM for the three components and the observed σ_{TF} s at $E_{\text{imp}} = 2, 8$, and 20 keV, the $\sigma_{\text{TF}}(E_{\text{imp}})$ curve without SCL is obtained as [Fig. 1(c)] :

$$\begin{aligned} \sigma_{\text{TF}}^{\text{preS}}(E_{\text{imp}}) &= V_{\text{F}}^{\text{Cu}}(E_{\text{imp}})\sigma_{\text{TF}}(2 \text{ keV}) + V_{\text{F}}^{\text{Li}_3\text{PO}_4}(E_{\text{imp}})\sigma_{\text{TF}}(8 \text{ keV}) \\ &+ V_{\text{F}}^{\text{LiCoO}_2}(E_{\text{imp}})\sigma_{\text{TF}}(20 \text{ keV}). \end{aligned} \quad (2)$$

A similar procedure provides the $\sigma_{\text{TF}}^{\text{preT}}(E_{\text{imp}})$ curve based on the TRIMSP simulation as also seen in Fig. 1(c).

Figure 2 shows the relationship between σ_{TF} and E_{imp} at 320, 300, 250, 200, and 150 K together with the predicted $\sigma_{\text{TF}}(E_{\text{imp}})$ curves based on the SRIM and TRIMSP simulations: i.e., both $\sigma_{\text{TF}}^{\text{preS}}(E_{\text{imp}})$ and $\sigma_{\text{TF}}^{\text{preT}}(E_{\text{imp}})$ curves. When SCL is formed, the Li content decreases in SCL. This leads to a smaller σ_{TF} in SCL than the prediction. However, it is highly unlikely to show a systematic discrepancy between the measured $\sigma_{\text{TF}}(E_{\text{imp}})$ and $\sigma_{\text{TF}}^{\text{preS}}(E_{\text{imp}})$ [or $\sigma_{\text{TF}}^{\text{preT}}(E_{\text{imp}})$] curves. In order to emphasize such discrepancy, Figs. 3 and 4 show the E_{imp} dependence of $\delta\sigma_{\text{TF}} (= \sigma_{\text{TF}} - \sigma_{\text{TF}}^{\text{preS}})$ and $(= \sigma_{\text{TF}} - \sigma_{\text{TF}}^{\text{preT}})$ for the two samples. The $\delta\sigma_{\text{TF}}(E_{\text{imp}})$ curves do not show a sudden decrease at the interface, at which $V_{\text{F}}^{\text{Li}_3\text{PO}_4} = V_{\text{F}}^{\text{LiCoO}_2}$ based on the two simulations [Figs. 1(a) and 1(b)]. This indicates the following two possibilities:

1) SCL does not exist or 2) the thickness of SCL is too thin to be resolvable even with LE μ^+ SR. Since the thickness of SCL is predicted to increase after a charge and discharge reaction [3], it would be a good idea to study the interface of the charged multilayer battery with LE μ^+ SR.

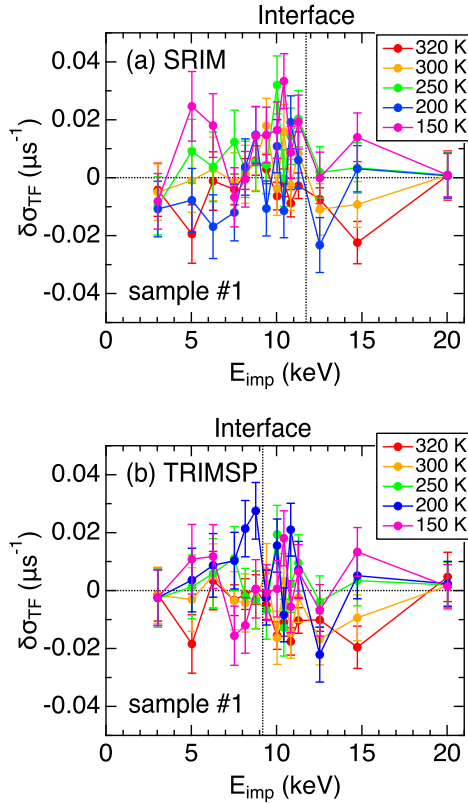


Figure 3. The E_{imp} dependence of the deviation of the measured σ_{TF} and predicted σ_{TF} ($\delta\sigma_{\text{TF}}$) with (a) SRIM and (b) TRIMSP for the sample #1. Vertical dotted lines labeled “Interface” represent the E_{imp} , at which the the volume fraction of Li_3PO_4 is equivalent to that of LiCoO_2 with the two simulations [see Figs. 1(a) and 1(b)].

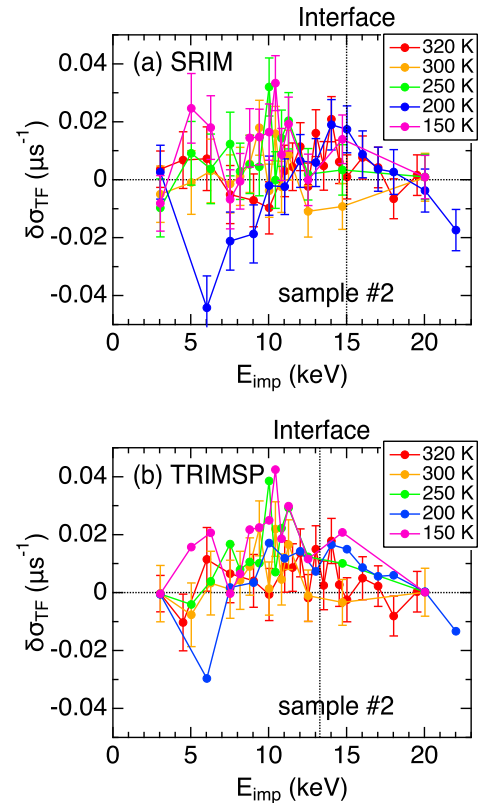


Figure 4. The E_{imp} dependence of the deviation of the measured σ_{TF} and predicted σ_{TF} ($\delta\sigma_{\text{TF}}$) with (a) SRIM and (b) TRIMSP for the sample #2. Vertical dotted lines labeled “Interface” represent the E_{imp} , at which the the volume fraction of Li_3PO_4 is equivalent to that of LiCoO_2 with the two simulations.

In order to study the dynamic behavior, we have also measured the ZF- and LF- μ^+ SR spectra for the Li_3PO_4 layer in the sample #1 with $E_{\text{imp}} = 8$ keV, at which $V_{\text{F}}^{\text{Li}_3\text{PO}_4} \sim 100\%$ predicted by SRIM (72% by TRIMSP) [see Fig. 1(a) and 1(b)]. The ZF- and LF- μ^+ SR spectra were fitted with a dynamic Gaussian Kubo-Toyabe function. Figure 5 shows the temperature dependencies of the field distribution width (Δ) and the field fluctuation rate (ν) in the Li_3PO_4 layer. While Δ is roughly temperature independent in the whole temperature range measured, ν increases with temperature above 200 K, indicating the Li^+ diffusion in the Li_3PO_4 solid electrolyte. This behavior is expected to stabilize the SCL at low temperatures, although such stabilization is most unlikely to be seen in Figs. 3 and 4.

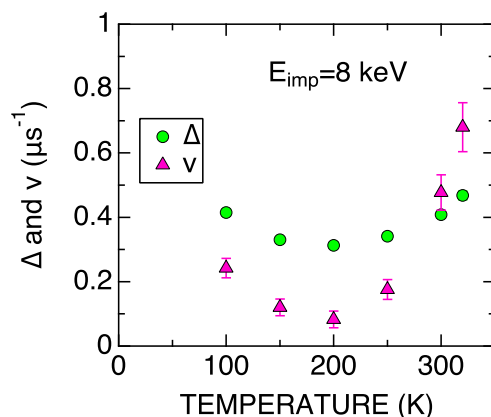


Figure 5. The temperature dependencies of the field distribution width (Δ) and field fluctuation rate (ν) in the sample #1 measured with $E_{\text{imp}} = 8$ keV, i.e., the center of the Li_3PO_4 layer.

4. Conclusion

Even with LE μ^+ SR, it was difficult to confirm the presence of SCL at the interface between the Li_3PO_4 electrolyte and LiCoO_2 electrode. For such purpose, it is important to make the interface at the vicinity of the surface in order to avoid the increase in the muon stopping distribution width with E_{imp} [17]. Moreover, it is preferable to study the interface of the sample after a charge and discharge reaction with LE μ^+ SR and/or an ultra-slow muon microscope in J-PARC.

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