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# Electrodes for High- $\kappa$ Molecular Crystal Antimony Trioxide Gate Dielectrics for 2D Electronics

Alok Ranjan,\* Lunjie Zeng, and Eva Olsson\*

Wafer-scale deposition of high- $\kappa$  gate dielectrics compatible with atomically thin van der Waals layered semiconductors (e.g., MoS<sub>2</sub>, WS<sub>2</sub>, WSe<sub>2</sub>) is urgently needed for practical applications of field effect transistors based on 2D materials. A study on a high- $\kappa$  molecular crystal antimony trioxide (Sb<sub>2</sub>O<sub>3</sub>) gate dielectric examined the role of electrode material on dielectric degradation and breakdown. It is demonstrated that the thin films of Sb<sub>2</sub>O<sub>3</sub> can be uniformly deposited on a wafer scale. The current–voltage (I–V) curves show tightly controlled distributions of both leakage current and breakdown voltage. Electrical measurements reveal that defects are generated gradually upon electrical stressing. The evaluation of degradation is based on charge trapping, stress-induced leakage current, and dielectric breakdown measurements. The breakdown voltage distribution follows a tight monomodal Weibull distribution suggesting a high quality of the film. Comparing Ti and Au as gate electrodes, both the breakdown field and the tunnel current are affected by the choice of electrode material. Transmission electron microscopy reveals that the chemistry at the electrode/Sb<sub>2</sub>O<sub>3</sub> interface plays an important role and that Ti scavenges oxygen from the Sb<sub>2</sub>O<sub>3</sub>, forming a defective oxide layer at the Ti/Sb<sub>2</sub>O<sub>3</sub> interface. For the Au electrode, this interfacial reaction is completely absent, improving the dielectric performance.

control its performance.<sup>[4,5]</sup> With the FETs reaching the physical limits of scaling, new materials and device architectures are actively explored. Atomically thin van der Waals layered materials (also called two dimensional (2D) material) have demonstrated an enormous potential to realize energy-efficient and ultrafast electronic and optical devices.<sup>[6–10]</sup> FETs based on the 2D semiconductors (e.g., MoS<sub>2</sub>, WS<sub>2</sub>, WSe<sub>2</sub>) have a higher intrinsic carrier mobility<sup>[11,12]</sup> compared to Si and they have also been shown to mitigate the short channel effects in ultra-scaled devices.<sup>[13,14]</sup> The successful integration of transistors based on the 2D materials in integrated circuits requires engineering the growth of high-quality 2D crystals on a wafer scale, understanding the physics of the semiconductor-dielectric interfaces, understanding the contact at the semiconductor-bulk metal interface, and addressing the reliability of the devices. One immediate challenge is to find a suitable gate dielectric, which could be seamlessly integrated with these

## 1. Introduction

There has been a continuous improvement in the performance of the field effect transistor (FET) in the last few decades.<sup>[1]</sup> Silicon has been a material of choice as a channel due to its modest carrier mobility and ability to form a high-quality semiconductor-dielectric interface.<sup>[2,3]</sup> The state-of-the-art gate-all-around nanosheet FET has been introduced at the 5 nm technology node, where only a handful of atoms in the channel

emerging 2D semiconductors.<sup>[15–17]</sup>

Hexagonal Boron Nitride<sup>[18]</sup> (h-BN) is an interesting material since it has been used as an atomically flat substrate,<sup>[19]</sup> an encapsulating layer,<sup>[12,20]</sup> and a gate dielectric in the prototype devices.<sup>[21–24]</sup> The h-BN forms a clean and defect-free van der Waals interface with the majority of the reported 2D semiconductors.<sup>[20]</sup> However, h-BN fails to meet the stringent requirements when 2D material-based transistors are scaled down.<sup>[16,17]</sup> This is because the gate dielectric thickness must be proportionately scaled as the transistor dimensions shrink in order to provide the capacitance effect. It has been found that the off-state leakage current becomes prohibitively higher in ultra-scaled devices, rendering h-BN unsuitable to be used as an effective gate dielectric. For example, the physical thickness of h-BN is  $\approx 1.3$  nm at the effective oxide thickness (EOT) of 1 nm and the leakage current density ( $J$ ) exceeds  $\approx 10$  A m<sup>-2</sup> (at 1 V) due to direct tunneling.<sup>[17]</sup> An alternate strategy is to use a high- $\kappa$  dielectric, akin to Si technology,<sup>[3]</sup> as a physically thick dielectric can be used for the sub-1 nm EOT-scaled FETs.

There have been efforts to grow the high- $\kappa$  gate dielectrics, e.g., HfO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, etc. directly on the 2D layered materials using atomic layer deposition (ALD).<sup>[25]</sup> The resulting dielectric films are inhomogeneous and also have poor coverage. This is because

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the surface of 2D materials is chemically inert to the ALD process, inhibiting the promotion of minimum nucleation sites required to grow a continuous thin film. Surface modification of the 2D materials through plasma<sup>[26]</sup> and ozone<sup>[27]</sup> treatment has shown improvements in the quality of the ALD-deposited dielectric films. However, exposure to either plasma or ozone also affects the crystallinity of the atomically thin 2D semiconductors and severely reduces their carrier mobility.

An interfacial seeding layer<sup>[28]</sup> deposited prior to ALD has been shown to promote the density of nucleation sites on the 2D semiconductors. These seeding layers typically have a low- $\kappa$  dielectric constant and therefore the overall dielectric constant achieved is then lower than that of the pristine ALD dielectric films. Alternatively, native oxides have been also grown directly on the 2D materials.<sup>[29–32]</sup> For example, the Bi<sub>2</sub>SeO<sub>5</sub> dielectric grown on the Bi<sub>2</sub>O<sub>2</sub>Se<sub>2</sub> semiconductor shows a semiconductor/dielectric interface that is atomically sharp.<sup>[29]</sup> This approach, however, is limited to a specific substrate at present, and a different strategy for the deposition of high- $\kappa$  gate dielectrics on 2D surfaces is required. The choice of the molecular crystal high- $\kappa$  ( $\kappa = 11.5$ ) antimony trioxide (Sb<sub>2</sub>O<sub>3</sub>) as a gate dielectric is promising<sup>[33,34]</sup> as Sb<sub>2</sub>O<sub>3</sub> can be grown directly on the surfaces of 2D crystals (e.g., MoS<sub>2</sub>, Graphene) using standard thermal evaporation deposition (STED).<sup>[33]</sup> The deposition process is scalable and the thickness of the Sb<sub>2</sub>O<sub>3</sub> films can be precisely controlled down to a nanometer thickness over a 3" wafer. Using Sb<sub>2</sub>O<sub>3</sub> as a top gate dielectric, prototype MoS<sub>2</sub> FET shows an on/off ratio of 10<sup>8</sup> and a record sub-threshold swing of 64 mV/decade at 300 K.<sup>[33]</sup>

In this work, we study the defect generation and evaluate the role of electrode material on the performance and dielectric breakdown of Sb<sub>2</sub>O<sub>3</sub> films by combining electrical measurements and nanoscale analysis using transmission electron microscopy (TEM). Comparing Ti and Au electrodes, we show that the choice of electrode affects the tunnel current and the dielectric breakdown strength of the Sb<sub>2</sub>O<sub>3</sub>. The results show that the leakage current increases by 3–4 orders of magnitude for the Ti electrode compared to the Au. By using an Au electrode, the breakdown field strength can be increased by  $\approx 1$  MV cm<sup>-1</sup> compared to Ti. The energy dispersive X-ray spectroscopy (EDXS) shows that the Ti scavenges oxygen from the Sb<sub>2</sub>O<sub>3</sub> and forms non-stoichiometric oxides and defects at the interface. The growth of the interfacial layer is absent for the Au electrode.

## 2. Results

### 2.1. Test Structure and Experimental Methods

The test structure consists of Ti/Sb<sub>2</sub>O<sub>3</sub>/n<sup>++</sup>Si and Au/Sb<sub>2</sub>O<sub>3</sub>/n<sup>++</sup>Si capacitors as schematically shown in **Figure 1a**. n<sup>++</sup>Si is used as a global bottom electrode as it provides an atomically flat surface for the deposition of Sb<sub>2</sub>O<sub>3</sub> films on 3" Si wafers. Capacitors of areas 50 × 50, 75 × 75, 100 × 100, and 150 × 150 μm<sup>2</sup> are fabricated and the details of Sb<sub>2</sub>O<sub>3</sub> film deposition, and the device fabrication can be found in the Experimental Section. The electrical measurements are carried out at room temperature using a semiconductor parameter analyzer connected to a probe station. This measurement setup serves two important purposes: i) enables the measurement of current

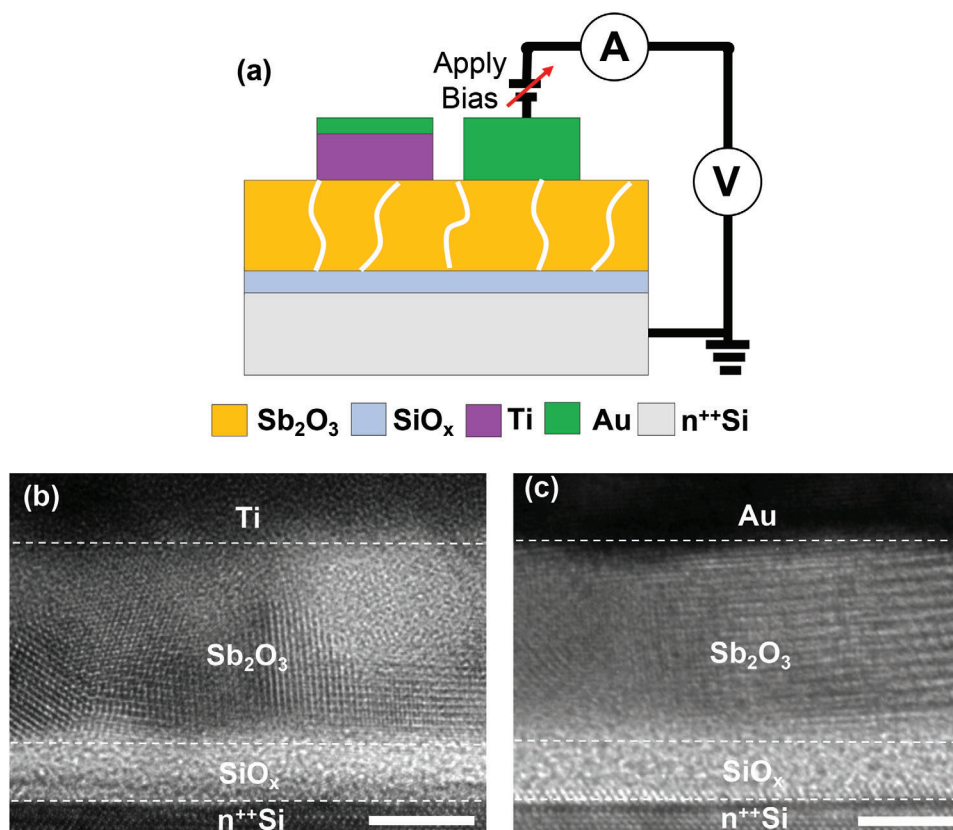
in the logarithmic range from pA to mA range and, ii) limits the current flow immediately when dielectric breakdown occurs. This is important for the study of dielectric breakdown since it provides access to different stages of dielectric degradation including charge trapping, stress-induced leakage current (SILC), and dielectric breakdown for a holistic electrical characterization.

We use transmission electron microscopy (TEM) to determine the film thickness and crystalline quality of the Sb<sub>2</sub>O<sub>3</sub> films in the prepared capacitors as shown in **Figure 1b,c**. The results show that the Sb<sub>2</sub>O<sub>3</sub> films are continuous, and the thickness of the film varies between  $\approx 7.5$  and 8.0 nm. The Sb<sub>2</sub>O<sub>3</sub> films are polycrystalline, and the size of nanocrystals is of the same order as the Sb<sub>2</sub>O<sub>3</sub> film thickness. A 3.0–3.3 nm thick interfacial SiO<sub>x</sub> layer is also present between the n<sup>++</sup>Si substrate and the Sb<sub>2</sub>O<sub>3</sub> film.

### 2.2. Time-Dependent Dielectric Breakdown and Random Telegraph Noise

We evaluate the integrity of the dielectric by carrying out time-dependent dielectric breakdown (TDDB) measurements. TDDB tests are standard dielectric reliability tests used to probe the temporal evolution of stress-induced defects and their impact on the tunneling current. In TDDB measurements, an electric field is applied, and the current is monitored as a function of time until breakdown occurs. The choice of applied electric field is arbitrary and typically ranges between 50% and 80% of the breakdown field, to accelerate the degradation in the dielectric and complete the measurements within a reasonable time. TDDB measurements on 150 × 150 μm<sup>2</sup> capacitors subjected to a bias of 1.5 V are shown in **Figures 2a** and **S1** (Supporting Information). The current increases progressively with time until breakdown occurs. Various stages of dielectric degradation are identifiable, including charge trapping, stress-induced leakage current (SILC), and multiple successive breakdown events.<sup>[35,36]</sup> The first stage is charge trapping where the tunnel current drops to  $\approx 15\%$  of the initial value. This initial reduction in current is due to pre-existing defects in the Sb<sub>2</sub>O<sub>3</sub>. The charge trapping component of the current can be better visualized by subtracting the current at  $t = 0$  sec from the overall TDDB current as shown in **Figure 2b**. This initial reduction in the current is linear and has also been observed for other gate dielectrics like SiO<sub>2</sub><sup>[37]</sup> and h-BN.<sup>[22]</sup> Following charge trapping, the SILC stage initiates and the tunnel current increases with time until breakdown occurs. It is considered that the continued electrical stress leads to the dilation of the created breakdown path and results in the formation of additional multiple breakdown paths.<sup>[36]</sup> The discrete jumps in the measured current that can be seen in **Figure 2a** are consistent with this assumption.

These pre-existing defects in Sb<sub>2</sub>O<sub>3</sub> also give rise to random telegraph noise (RTN),<sup>[38]</sup> where examples are shown in **Figure 2c–e**. RTN in gate dielectrics is associated with the capture and emission of charge carriers at the individual defect sites which manifests in the fluctuations of tunnel current between well-defined levels at random time intervals.<sup>[39,40]</sup> The RTN is observed when an electron is captured at the defect site as it increases the local potential barrier for the incoming charge carriers to tunnel and hence leads to a reduction in the current. The current recovers its original value when the captured



**Figure 1.** a) Schematic showing the cross-section of the capacitor test structure and the electrical connections for the charge transport measurements. Capacitors with both Ti and Au as top electrodes are fabricated. The capacitor areas are  $50 \times 50$ ,  $75 \times 75$ ,  $100 \times 100$ , and  $150 \times 150 \mu\text{m}^2$ . The bias is applied to the top electrode and  $n^{++}\text{Si}$  is grounded during the electrical measurements. b) TEM micrograph showing a cross-section of the as-prepared capacitor with Ti. c) TEM micrograph showing a cross-section of the as-prepared capacitor with Au electrode. The Ti is capped with a thin layer of Au to prevent oxidation. The  $\text{Sb}_2\text{O}_3$  film is continuous and has a thickness of 7.5–8.0 nm. The  $\text{Sb}_2\text{O}_3$  film is polycrystalline as evidenced by the presence of lattice fringes observed in (b) and (c). An interfacial  $\text{SiO}_x$  layer of thickness 3.0–3.3 nm is also present. The scale bars in (b) and (c) are 5 nm.

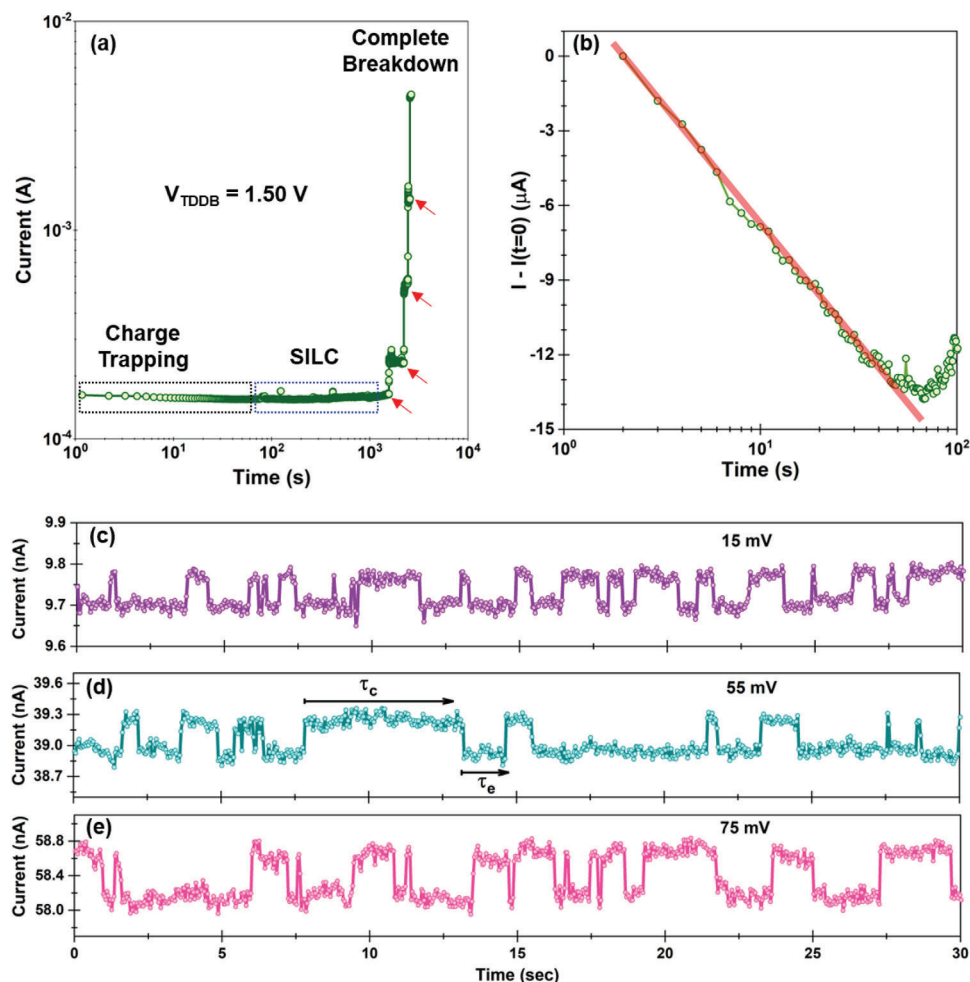
electron is emitted from the defect site. For each current fluctuation, the time for which the current remains high and low is known as the time to capture ( $\tau_c$ ) and time to emit ( $\tau_e$ ) respectively, as shown in Figure 2d. For the measurement of RTN, a small bias (typically < 20% of the breakdown field) is applied and the tunnel current is measured as a function of time. RTN is sensitive to the applied electric field and hence the bias is changed only in small steps to obtain statistical trends.

RTNs are readily observed across capacitors with Ti electrodes. For the data shown in Figure 2c–e, a 2-level RTN is consistently observed for an applied voltage range of 25–75 mV. Here, only a small portion of the data is plotted for brevity and the complete RTN spectra are shown in Figure S2 (Supporting Information). Both  $\tau_c$  and  $\tau_e$  are extracted from each RTN spectrum and analyzed further. Both  $\tau_c$  and  $\tau_e$  show exponential distributions for all the RTNs as shown by the histogram in Figures S3 and S4 (Supporting Information). The exponential distribution of the time constants has also been observed for RTNs in other gate dielectrics.<sup>[38,39,41]</sup> Interestingly, we also observe that when the time constants ( $\tau_c$  and  $\tau_e$ ) for all the RTN data are plotted on the same exponential plot, all the distributions overlap as shown in Figure S5 (Supporting Information). This overlap of the distributions of all six RTN spectra highlights that a single defect

is associated with all the RTNs measured here.<sup>[42]</sup> Additionally, multi-level RTN is also observed, and an example is shown in Figure S6 (Supporting Information).

### 2.3. Role of Electrodes on Dielectric Breakdown

In this section, we discuss the role of electrodes on the dielectric breakdown in  $\text{Sb}_2\text{O}_3$ . Tunnel current and breakdown field ( $E_{\text{BD}}$ ) are two critical parameters affecting the reliability of a gate dielectric and hence we evaluate these for  $\text{Sb}_2\text{O}_3$  in this section. The tunnel current measured for capacitors of area  $50 \times 50 \mu\text{m}^2$  with Ti electrodes is plotted in Figure S7 (Supporting Information). Here, the tunnel current is measured by applying a ramp voltage stress (RVS) between –250 and +250 mV. The measurements are also carried out on capacitors of areas  $75 \times 75$  and  $100 \times 100 \mu\text{m}^2$  and the current density ( $J$ ) is obtained by dividing the measured tunnel current by the capacitor area. The current density versus voltage ( $J$ – $V$ ) plot for all three capacitor areas is shown in Figure S8 (Supporting Information). The  $J$ – $V$  plots for all the capacitors overlap suggesting that the tunnel current scales with the area of the capacitor and no bulk structural defects, such as pinholes, are present in the  $\text{Sb}_2\text{O}_3$  film. This agrees with the TEM analysis in



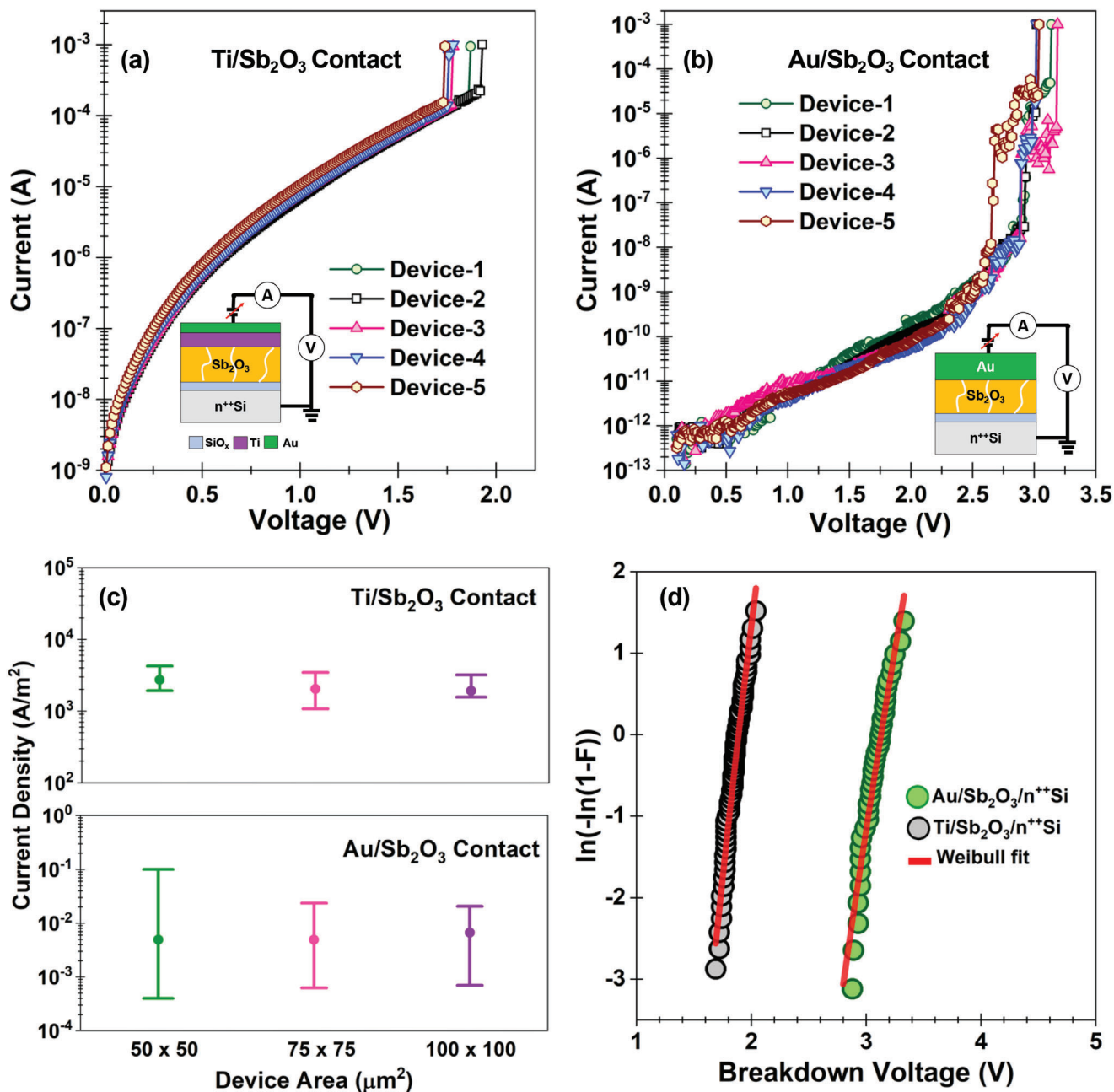
**Figure 2.** a) Time-dependent dielectric breakdown (TDDB) measurements showing the different stages of dielectric degradation including charge trapping, stress-induced leakage current (SILC), and breakdown in Ti/Sb<sub>2</sub>O<sub>3</sub>/n<sup>++</sup>Si capacitors. Multiple breakdowns can be observed as indicated by the red arrows. b) The charge trapping component can be extracted from the TDDB current (i.e.,  $I(t)$ ) by subtracting the time zero current (i.e.,  $I(t = 0)$ ). The reduction in the charge-trapping current fits a linear curve on the log-normal scale. c–e) Random telegraph noise originating from the pre-existing defects in Sb<sub>2</sub>O<sub>3</sub>. The complete RTN spectra are shown in Figure S2 (Supporting Information). The capture ( $\tau_c$ ) and emission time ( $\tau_e$ ) are labeled in (d).

Figure 1b,c which shows that the Sb<sub>2</sub>O<sub>3</sub> film is continuous with a uniform thickness.

We now evaluate the breakdown voltage of the capacitors by applying an RVS until breakdown occurs and a current compliance of  $\approx 1$  mA is reached. The representative breakdown current–voltage ( $I$ – $V$ ) curves measured on five different capacitors of  $50 \times 50 \mu\text{m}^2$  area are shown in Figure 3a. A sudden increase in current is observed at the onset of breakdown. The post-breakdown  $I$ – $V$  in Figure S9 (Supporting Information) shows that the resistance is  $\approx 7.5$  k $\Omega$ . The low resistance after the breakdown and the ohmic behavior of the post-breakdown  $I$ – $V$  curve suggest that a metallic conducting filament has been formed. In our analysis, we define the breakdown voltage ( $V_{\text{BD}}$ ) as the voltage at which the current compliance reaches  $\approx 1$  mA and this is used to extract the breakdown field ( $E_{\text{BD}}$ ). The average value of the  $V_{\text{BD}}$  of Sb<sub>2</sub>O<sub>3</sub> with Ti electrode is 1.9 V as obtained from the Weibull analysis shown in Figure 3d.  $E_{\text{BD}}$  is obtained by dividing  $V_{\text{BD}}$  by the dielectric thickness ( $t_{\text{dielectric}}$ ) and

we use the total dielectric thickness of  $\approx 11.3$  nm (i.e., 8.0 nm Sb<sub>2</sub>O<sub>3</sub> + 3.3 nm SiO<sub>x</sub>) by considering both the Sb<sub>2</sub>O<sub>3</sub> thickness and the interfacial SiO<sub>x</sub> layer. It should be noted that this represents a worst-case scenario and provides a lower bound on the estimates of  $E_{\text{BD}}$  as interfacial oxides are defective. The calculated  $E_{\text{BD}}$  for Ti electrodes is  $\approx 1.7$  MV cm<sup>-1</sup> and we find that the values are lower by  $\approx 1.0$  MV cm<sup>-1</sup> compared to InAu electrodes.<sup>[33]</sup>

To investigate the origin of this reduction in the breakdown field, we carried out a breakdown analysis on capacitors with Au electrodes. The representative breakdown  $I$ – $V$  curves on  $50 \times 50 \mu\text{m}^2$  capacitors are shown in Figure 3b. There are two distinct observations: a) there is a large reduction in the tunnel current, and b) the breakdown voltage is increased for capacitors with Au electrodes. For the quantitative analysis, the current density for capacitors with both Ti and Au electrodes for three different areas of  $50 \times 50$ ,  $75 \times 75$ , and  $100 \times 100 \mu\text{m}^2$  at 1 V is plotted in Figure 3c. We observe that while the current density distribution

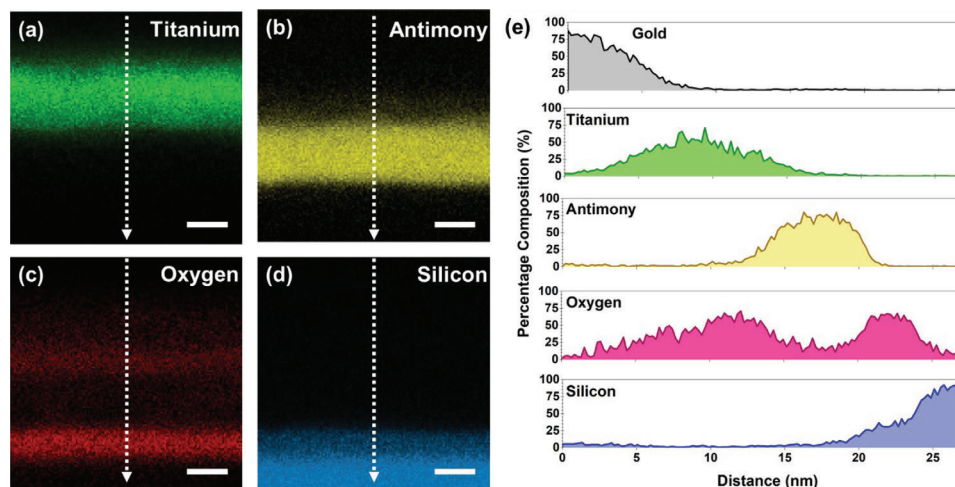


**Figure 3.** Breakdown current–voltage ( $I$ – $V$ ) curves measured on  $\text{Sb}_2\text{O}_3$  film with a) a Ti and b) an Au electrode. c) The plot of current density for the three different capacitor areas of  $50 \times 50$ ,  $75 \times 75$ , and  $100 \times 100 \mu\text{m}^2$  for Ti and Au electrodes measured at a bias of 1 V. At least 10 devices have been measured for each capacitor area to obtain the boxplot shown in (c). d) Plot of breakdown voltage for both Ti and Au electrodes on  $\text{Sb}_2\text{O}_3$  on a Weibull scale. The breakdown is defined as the voltage at which the current reaches a compliance of  $\approx 1$  mA. At least 40 capacitors have been measured for the Ti and the Au electrodes each to obtain the statistical plot shown in (d). The areas of the capacitors in (a), (b), and (d) are  $50 \times 50 \mu\text{m}^2$ .

is narrow for the Ti electrodes, their absolute value reduces by at least four orders of magnitude for Au electrodes.

$V_{\text{BD}}$  values for  $\text{Sb}_2\text{O}_3$  capacitors with both Ti and Au electrodes are extracted following the procedure discussed above. Following the standard failure reliability approaches,  $V_{\text{BD}}$  data are plotted on a Weibull scale<sup>[43,44]</sup> as shown in Figure 3d. We note that breakdown distribution for capacitors with both Ti and Au electrodes follows a linear Weibull model. The Weibull slope ( $\beta$ ) is an impor-

tant fitting parameter that gives an estimation about the spread in the breakdown voltage. Generally,  $\beta < 1$  indicates that the breakdown is caused by extrinsic defects.<sup>[45]</sup> Therefore,  $\beta > 1$  indicates that the intrinsic defects determine the breakdown. A larger value of  $\beta$  indicates a higher uniformity in the thickness of the dielectric film. The extracted values of  $\beta$  for both Ti and Au electrodes are 27.5 and 24.5 respectively. We also note that the values of  $\beta$  in the  $\text{Sb}_2\text{O}_3$  are comparable to a VLSI quality 3 nm thick  $\text{SiO}_2$



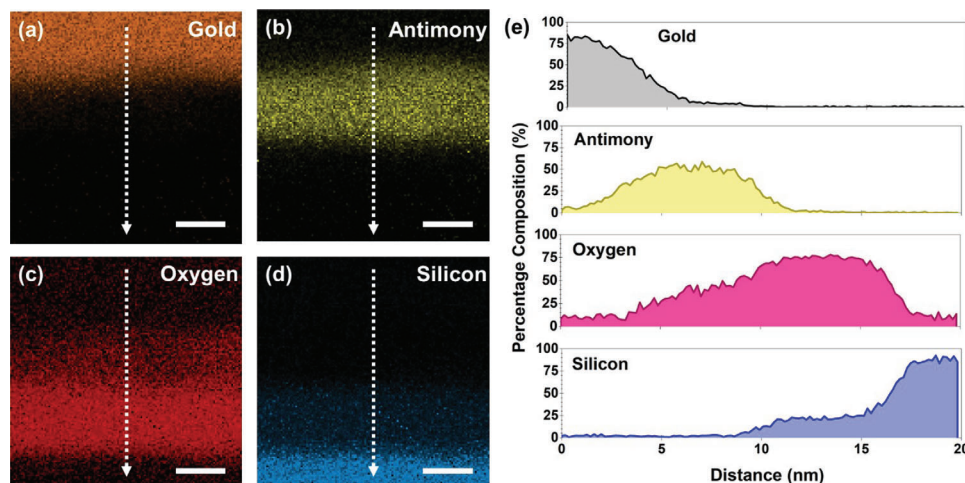
**Figure 4.** EDXS chemical analysis of the Ti/Sb<sub>2</sub>O<sub>3</sub>/n<sup>++</sup>Si capacitors: a–d) EDXS elemental maps showing the elemental distribution of Ti, Sb, O, and Si along the white arrow. Ti is capped with Au to prevent oxidation from the ambient. The Au elemental map is not shown here for brevity. e) Line profiles across the white arrows in (a–d) showing the atomic percentage of all elements. The line profiles are obtained by averaging across 20 pixels to show the average trends. Here, each pixel is 0.18 nm wide. It should be noted that there is an accumulation of oxygen at the Ti/Sb<sub>2</sub>O<sub>3</sub> interface and a simultaneous decrease in the oxygen concentration of the Sb<sub>2</sub>O<sub>3</sub> layer. The scale bars in (a–d) are 5 nm.

film.<sup>[46]</sup> The mean values of breakdown voltage ( $\tau_{BD}$ ) obtained from the Weibull fitting for Ti and Au electrodes are  $\approx 1.90$  and  $\approx 3.13$  V, respectively. The measured  $E_{BD}$  for Au electrode is  $\approx 2.76$  MV cm<sup>-1</sup> (compared to  $\approx 1.68$  MV cm<sup>-1</sup> for Ti electrode) and we discuss the origin of the differences in the electrical behavior below.

## 2.4. Chemical Analysis of Ti and Au Electrodes with Sb<sub>2</sub>O<sub>3</sub>

We use scanning transmission electron microscopy (STEM) along with electron energy loss spectroscopy (EELS) and EDXS to decipher the origin of the differences observed for Ti and Au electrodes. Cross-section TEM lamellae are prepared from

both Ti/Sb<sub>2</sub>O<sub>3</sub>/n<sup>++</sup>Si and Au/Sb<sub>2</sub>O<sub>3</sub>/n<sup>++</sup>Si capacitors prior to electrical stress and the results of the analyses are shown in **Figures 4** and **5**. EDXS data of Ti, Sb, O, and Si are plotted in **Figure 4a–d** which shows the distribution of the elements in the different layers. A line profile showing the elemental distribution of Ti, Sb, O, and Si in a capacitor with a Ti electrode is shown in **Figure 4e** and we find that there is a significant redistribution of oxygen both at the Ti/Sb<sub>2</sub>O<sub>3</sub> interface and in the Sb<sub>2</sub>O<sub>3</sub> bulk. The quantitative analysis of the EDXS data in **Figure 4e** shows that the ratio of oxygen to antimony content at the Sb<sub>2</sub>O<sub>3</sub> bulk region of the film is drastically reduced to 0.26 compared to the ideal stoichiometric Sb<sub>2</sub>O<sub>3</sub> (which has a ratio of 1.5). The concentration profiles in **Figure 4e** show that there is a formation of a TiO<sub>x</sub> layer at the Ti/Sb<sub>2</sub>O<sub>3</sub> interface. During the



**Figure 5.** EDXS chemical analysis of the Au/Sb<sub>2</sub>O<sub>3</sub>/n<sup>++</sup>Si capacitors: a–d) EDXS elemental maps show the elemental distribution of Au, Sb, O, and Si along the white arrow. e) Line profiles across the white arrows in (a–d) showing the atomic percentage of all elements. The line profiles are obtained by averaging across 20 pixels to show the average trends. Here, each pixel is 0.20 nm wide. Unlike Ti electrodes, there is no accumulation of oxygen at the Au/Sb<sub>2</sub>O<sub>3</sub> interface. The scale bars in (a–d) are 5 nm.

formation of the  $\text{TiO}_x$  layer, the  $\text{Sb}_2\text{O}_3$  layer loses oxygen leading to the formation of a sub-stoichiometric  $\text{Sb}_2\text{O}_3$ , deficient in oxygen.

Additional evidence for the formation of  $\text{TiO}_x$  is also provided by the EELS analysis as shown in Figure S10 (Supporting Information). The onset of the core-loss EELS edge and the energy loss near edge fine structure (ELNES) are sensitive to the chemical environment in the material, and they can be used as an indicator to probe if and how the oxygen is chemically bonded to Ti. We observe that the Ti L-edge EELS signal at the  $\text{Ti/Sb}_2\text{O}_3$  interface shows a positive shift in the onset of the edge by  $\approx 1.5$  eV when compared to the bulk Ti. The Ti-L ELNES is consistent with titanium oxides ( $\text{TiO}_x$ , where  $x < 2$ ).<sup>[47,48]</sup> It is challenging to quantify the value of  $x$  as there is a gradual change in the oxygen concentration in the Ti film, with the highest oxygen concentration at the  $\text{Ti/Sb}_2\text{O}_3$  interface as shown in Figure 4e.

A similar elemental analysis for capacitors with Au electrodes is shown in Figure 5a–d and the distribution of elements agrees with the fabrication process. A line profile showing the distributions of Au, Sb, O, and Si is plotted in Figure 5e. The analysis of the EDXS data shows that the elemental ratio of oxygen to antimony content at the  $\text{Sb}_2\text{O}_3$  bulk region of the film is 0.79. We note that the oxygen content in the  $\text{Sb}_2\text{O}_3$  bulk with the Au electrode is about three times higher as compared to the Ti electrode. However, the  $\text{Sb}_2\text{O}_3$  films with Au electrodes are also non-stoichiometric and oxygen-deficient.

## 2.5. Discussion

The general trend of the time-dependent dielectric degradation in  $\text{Sb}_2\text{O}_3$  shown in Figure 2a is similar to other types of gate dielectrics (e.g.,  $\text{SiO}_2$ ,  $\text{HfO}_2$ , and h-BN). The various stages of dielectric degradation including charge trapping, SILC, and breakdown are clearly identifiable. We find that the oxygen-deficient  $\text{Sb}_2\text{O}_3$  film causes charge trapping, and the effect is readily visible in the initial stages of TDDB where the tunnel current can drop and decrease can be up to  $\approx 6\%$  of the initial value. This charge-trapping effect is especially dominant for the capacitors where Ti is used as a top electrode. The Ti scavenges oxygen from  $\text{Sb}_2\text{O}_3$  forming a defective interfacial layer. These pre-existing defects in the oxygen vacancy-rich  $\text{Sb}_2\text{O}_3$  film also lead to RTN current fluctuations. Both 2-level and multi-level RTNs are readily observed in the  $\text{Sb}_2\text{O}_3$ . The RTN defects in  $\text{Sb}_2\text{O}_3$  are stable at room temperature and hence long-term RTN spectra can be measured. Although this RTN is a reliability challenge for the FETs, these oxygen-deficient  $\text{Sb}_2\text{O}_3$  layers have potential for applications in cryptography as sources of RTN generators since they offer the possibility of controlled tunability of oxygen vacancies.<sup>[49]</sup>

A critical finding of this study is that the interfacial reaction at the dielectric/electrode interface plays a crucial role in the overall performance of the  $\text{Sb}_2\text{O}_3$ . We demonstrate this by comparing the two commonly used electrodes Ti and Au. The tunnel current is significantly higher for Ti as compared to Au. Given that the  $\text{Sb}_2\text{O}_3$  layer thickness is the same for both the Ti and Au electrodes, the 3–4 orders of increase in the tunnel current for Ti electrodes cannot be accounted for considering the lower work function of Ti ( $\phi_{\text{Ti}} = 4.33$  eV) compared to Au ( $\phi_{\text{Au}} = 5.20$  eV). Critical insights emerge from the cross-sectional TEM analysis of the fab-

ricated capacitors which shows that there is a significantly higher concentration of oxygen present at the  $\text{Ti/Sb}_2\text{O}_3$  interfaces. This suggests the oxidation of Ti and the formation of an oxygen-deficient non-stoichiometric  $\text{Sb}_2\text{O}_3$ . This is further corroborated by the EELS analysis of Ti which shows that the onset for the L-edge is delayed at the interface as compared to the metallic Ti electrode, indicating the formation of Ti–O bonds at the interface. Ti-L ELNES further confirms the formation of a  $\text{TiO}_x$  phase at the interface. In contrast, we do not observe the formation of any interfacial oxide at the  $\text{Au/Sb}_2\text{O}_3$  interface. This absence of the interfacial oxide formation at the Au electrode also increases the breakdown field of  $\text{Sb}_2\text{O}_3$  to  $2.7$  MV  $\text{cm}^{-1}$ . This value agrees with the breakdown field strength of  $\text{Sb}_2\text{O}_3$  of  $2.7$  MV  $\text{cm}^{-1}$  measured previously using an InAu electrode.<sup>[33]</sup> We note that the deposited  $\text{Sb}_2\text{O}_3$  films in all the cases are non-stoichiometric and oxygen deficient, and the improvement in the deposition process has the potential to increase the breakdown field strength.

This work also addresses another key practical aspect for the successful integration of the high- $\kappa$   $\text{Sb}_2\text{O}_3$  films into the fabrication process. We demonstrate that a continuous, homogeneous, and defect-free (e.g., with any pinholes)  $\text{Sb}_2\text{O}_3$  film can be deposited on a wafer scale for the fabrication of devices. The electrical measurements carried out on the capacitors show that tunnel current variations from device to device have a narrow distribution as shown in Figure S7 (Supporting Information). Given that the tunnel current is highly sensitive to the dielectric thickness, it can be also used as an indirect indication of the uniformity of the  $\text{Sb}_2\text{O}_3$  thickness across the 3" inch Si wafer. Direct evidence of the uniform thickness is also provided by the TEM measurements in Figure 1b,c which shows a uniformity of the  $\text{Sb}_2\text{O}_3$  layer.

These initial results are promising for further consideration of  $\text{Sb}_2\text{O}_3$  as a gate dielectric for 2D material-based transistors. However, there are many questions related to the material growth, metal/dielectric interfaces, physics of charge transport, and dielectric breakdown that remain to be understood. For example, it is not clear at present why the  $\text{Sb}_2\text{O}_3$  film is polycrystalline, and non-stoichiometric, and how the crystallinity is affected by the growth substrate. It would also be important to further investigate the metal/dielectric interfaces for other commonly used gate electrodes. Future studies should also focus on understanding the atomic structure and improving the  $\text{Sb}_2\text{O}_3$  deposition process to obtain stoichiometric films. Considering the different process steps in the fabrication process, the thermal stability of  $\text{Sb}_2\text{O}_3$  is another crucial aspect and would be subject to future work.

Optimizing the electrical contacts presents a significant challenge when introducing new materials in the devices. For example, there have been significant efforts undertaken to understand and engineer the 2D semiconductor-metal contact interface to meet the technology requirements.<sup>[50]</sup> The challenge with the new emerging gate dielectrics is to identify a suitable contact metal that is chemically inert as well as relevant for applications. Insights gained from the previous works on 2D semiconductor-metal contact interfaces can be applied to optimize the contacts for the emerging gate dielectrics.<sup>[51]</sup> It is important to point out that the use of noble elements like Au or InAu as a contact material would have limited application for large-scale technology due to requirements from semiconductor fabrication. It is important to explore alternative pathways for improving the contact property. One promising approach is to use a diffusion barrier



layer (e.g., TiN) between the dielectric and the contact layer, which could prevent the chemical reaction from taking place at the dielectric/metal interface.

### 3. Conclusion

In this work, we have investigated the defect generation, degradation, and breakdown in high- $\kappa$  molecular crystal antimony trioxide ( $\text{Sb}_2\text{O}_3$ ) gate dielectrics. Electrical measurements show that the  $\text{Sb}_2\text{O}_3$  undergoes various stages of degradation including charge trapping and stress-induced leakage current prior to dielectric breakdown. The pre-existing atomic defects in  $\text{Sb}_2\text{O}_3$  give rise to random telegraph noise and both two-level and multi-level RTNs are observed. We show that the choice of electrode material is critical for the realization of the full potential of  $\text{Sb}_2\text{O}_3$  as a gate dielectric. Using Ti and Au as electrodes, we show that the chemistry at the  $\text{Sb}_2\text{O}_3$ /electrode interface dictates the tunnel current and the dielectric breakdown strength. EDXS and EELS analysis confirm that Ti scavenges oxygen from  $\text{Sb}_2\text{O}_3$ . This results in the formation of both an oxygen-deficient  $\text{Sb}_2\text{O}_3$  oxide at the interface as well as a defective  $\text{TiO}_x$  layer. This non-stoichiometric  $\text{TiO}_x/\text{Sb}_2\text{O}_{3-x}$  bi-layer has a high leakage current and a reduced breakdown voltage. This interfacial reaction is absent when Au is used as an electrode material highlighting the importance of selecting appropriate electrode materials for transistor applications.

### 4. Experimental Section

***Sb<sub>2</sub>O<sub>3</sub> Thin Film Deposition:*** A total of 3" n<sup>++</sup>Si wafers ( $\rho < 0.025 \Omega\cdot\text{cm}$ ) were pre-cleaned in acetone and IPA to remove surface contaminants. The wafers were immersed in hydrofluoric acid for 30 s to remove any native oxide. The  $\text{Sb}_2\text{O}_3$  film was deposited using standard thermal evaporation (Nexdep, Angstrom Engineering) in a high vacuum ( $10^{-6}$  Torr) following the procedure described in previously published work.<sup>[33]</sup> The thickness of the  $\text{Sb}_2\text{O}_3$  film was precisely controlled using an in situ crystal quartz monitor.

***Device Fabrication:*** Small pieces of  $\approx 2 \text{ cm} \times 2 \text{ cm}$  were diced from the 3" wafers and the top electrodes were patterned by electron beam lithography (EBL). Briefly, a 200 nm thick PMMA A4 resist was spin-coated on the sample at 4,000 rpm for 60 s and the sample was baked at 180 °C for 5 min prior to lithography. EBL was carried out using a JEOL JBX 9300FS at an acceleration voltage of 100 kV. The patterns were developed in MIBK-IPA (2:1) for 2.20 min, cleaned in IPA for 30 s, and dried in  $\text{N}_2$ . The top metal electrode was deposited using Kurt J. Lesker's Physical Vapor Deposition System under high vacuum conditions at room temperature. For bilayer electrodes (e.g., Ti/Au), metal layers were deposited without breaking the vacuum to avoid any oxidation. The lift-off was carried out in acetone for  $\approx 20$  min at 50 °C. The sample was cleaned in IPA and vacuum dried for  $\approx 3\text{--}4$  h prior to any measurements.

***Electrical Measurements:*** Electrical measurements were carried out using an Agilent B1400 semiconductor parameter analyzer connected to a probe station. The probe station was equipped with four independent source measurement units and the base noise for the current measurements was  $\approx 1$  pA. During all the measurements, bias was applied to the top electrode, and the bottom n<sup>++</sup>Si substrate was electrically grounded. All the measurements were carried out at room temperature. For RVS measurements, a ramp rate of  $0.4 \text{ V s}^{-1}$  was used and for RTN measurements a sampling rate of  $500 \text{ points s}^{-1}$  was used.

***Focused Ion Beam Sample Preparation:*** Cross-section samples for the TEM analysis were prepared using a FEI Versa 3D FIB-SEM. A few  $\mu\text{m}$  thick Pt protection layer was first deposited using the electron beam and

subsequently using the Ga ion beam. Lift-out was carried out to prepare a  $\approx 5 \mu\text{m}$  long TEM lamella. Final milling was carried out at an accelerating voltage of 5 and 2 kV respectively, to diminish amorphous layers and beam damage on the sample surface.

***Transmission Electron Microscopy:*** TEM measurements were carried out at 300 kV using a FEI Titan microscope. Scanning TEM (STEM) measurements were carried out at 200 kV on a JEOL Monochromated ARM200F microscope equipped with a CEOS GmbH probe corrector, a double silicon drift detector (SDD) for energy dispersive X-ray spectroscopy (EDXS), and a Gatan Imaging Filter Continuum for electron energy loss spectroscopy (EELS).

***Quantification of the EDXS data:*** Given the overlap of the O and Sb EELS edges, EDXS was used for the chemical analysis in Figures 4 and 5. EDXS composition quantification was carried out using the Gatan Microscopy Suite (GMS 3.40) software, with background subtraction using the Kramers model, theoretical K-factor, Casnati cross-section model, and thickness correction. The thickness was estimated using EELS data and the log ratio (relative) routine available in GMS. The result from the analysis is obtained in units of inelastic mean free path (IMFP) of the electrons,  $t/\lambda$ . Here,  $t$  is the sample thickness, and the  $\lambda$  is the IMFP of electrons in the material. The IMFP of the electrons measured in the Si substrate, at the region of the sample where the EDXS data were acquired, varies between 0.64 and 0.88 (mean = 0.76) and 1.25–1.58 (mean = 1.42) for Au and Ti electrodes respectively. Considering that the IMFP of electrons at 200 keV in Si is  $\approx 145 \text{ nm}$ ,<sup>[52]</sup> the estimated mean thickness of the TEM lamellae with Au and Ti electrodes are  $\approx 110$  and  $\approx 206 \text{ nm}$  respectively. Electron beam broadening can affect the quantitative analysis of the EDXS data at the interfaces. For the quantitative analysis of  $\text{Sb}_2\text{O}_3$  films, therefore selected the regions where the EDXS signal is not affected by the beam broadening and the average values are provided.

### Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

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### Conflict of Interest

The authors declare no conflict of interest.

### Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

### Keywords

2D materials, gate dielectric, high- $\kappa$ , molecular crystal, reliability,  $\text{Sb}_2\text{O}_3$

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- [1] R. K. Ratnesh, A. Goel, G. Kaushik, H. Garg, Chandan, M. Singh, B. Prasad, *Mater. Sci. Semicond. Process.* **2021**, *134*, 106002.
- [2] S. M. Sze, K. K. Ng, *Physics of Semiconductor Devices*, 3<sup>rd</sup> Edition, John Wiley & Sons, Hoboken, NJ, **2006**.
- [3] H. Wong, H. Iwai, *Microelectron. Eng.* **2015**, *138*, 57.
- [4] S. Mukesh, J. Zhang, *Electronics* **2022**, *11*, <https://www.mdpi.com/2079-9292/11/21/3589>.
- [5] N. Loubet, T. Hook, P. Montanini, C. W. Yeung, S. Kanakasabapathy, M. Guillom, T. Yamashita, J. Zhang, X. Miao, J. Wang, A. Young, R. Chao, M. Kang, Z. Liu, S. Fan, B. Hamieh, S. Sieg, Y. Mignot, W. Xu, S. C. Seo, J. Yoo, S. Mochizuki, M. Sankarapandian, O. Kwon, A. Carr, A. Greene, Y. Park, J. Frougier, R. Galatage, R. Bao, et al., in 2017 Symposium on VLSI Technology, **2017**, pp. T230–T231, <https://ieeexplore.ieee.org/document/7998183>.
- [6] K. S. Novoselov, A. Mishchenko, A. Carvalho, A. H. Castro Neto, *Science* **2016**, *353*, aac9439.
- [7] F. Hayee, L. Yu, J. L. Zhang, C. J. Ciccarino, M. Nguyen, A. F. Marshall, I. Aharonovich, J. Vučković, P. Narang, T. F. Heinz, J. A. Dionne, *Nat. Mater.* **2020**, *19*, 534.
- [8] A. S. Mayorov, R. V. Gorbachev, S. V. Morozov, L. Britnell, R. Jalil, L. A. Ponomarenko, P. Blake, K. S. Novoselov, K. Watanabe, T. Taniguchi, A. K. Geim, *Nano Lett.* **2011**, *11*, 2396.
- [9] L. Wang, I. Meric, P. Y. Huang, Q. Gao, Y. Gao, H. Tran, T. Taniguchi, K. Watanabe, L. M. Campos, D. A. Muller, J. Guo, P. Kim, J. Hone, K. L. Shepard, C. R. Dean, *Science* **2013**, *342*, 614.
- [10] Y. Liu, X. Duan, H.-J. Shin, S. Park, Y. Huang, X. Duan, *Nature* **2021**, *591*, 43.
- [11] S. Das, A. Sebastian, E. Pop, C. J. McClellan, A. D. Franklin, T. Grasser, T. Knobloch, Y. Illarionov, A. V. Penumatcha, J. Appenzeller, Z. Chen, W. Zhu, I. Asselberghs, L.-J. Li, U. E. Avci, N. Bhat, T. D. Anthopoulos, R. Singh, *Nat. Electron.* **2021**, *4*, 786.
- [12] G.-H. Lee, X. Cui, Y. D. Kim, G. Arefe, X. Zhang, C.-H. Lee, F. Ye, K. Watanabe, T. Taniguchi, P. Kim, J. Hone, *ACS Nano* **2015**, *9*, 7019.
- [13] K. Xu, D. Chen, F. Yang, Z. Wang, L. Yin, F. Wang, R. Cheng, K. Liu, J. Xiong, Q. Liu, J. He, *Nano Lett.* **2017**, *17*, 1065.
- [14] H. Liu, A. T. Neal, P. D. Ye, *ACS Nano* **2012**, *6*, 8563.
- [15] F. Palumbo, C. Wen, S. Lombardo, S. Pazos, F. Aguirre, M. Eizenberg, F. Hui, M. Lanza, *Adv. Funct. Mater.* **2020**, *30*, 1900657.
- [16] Y. Y. Illarionov, T. Knobloch, M. Jech, M. Lanza, D. Akinwande, M. I. Vexler, T. Mueller, M. C. Lemme, G. Fiori, F. Schwierz, *Nat. Commun.* **2020**, *11*, 3385.
- [17] Y. C. Lin, C. M. Lin, H. Y. Chen, S. Vaziri, X. Bao, W. Y. Woon, H. Wang, S. S. Liao, *IEEE Trans. Electron Devices* **2023**, *70*, 1454.
- [18] K. Watanabe, T. Taniguchi, H. Kanda, *Nat. Mater.* **2004**, *3*, 404.
- [19] C. R. Dean, A. F. Young, I. Meric, C. Lee, L. Wang, S. Sorgenfrei, K. Watanabe, T. Taniguchi, P. Kim, K. L. Shepard, J. Hone, *Nat. Nanotechnol.* **2010**, *5*, 722.
- [20] A. V. Kretinin, Y. Cao, J. S. Tu, G. L. Yu, R. Jalil, K. S. Novoselov, S. J. Haigh, A. Gholinia, A. Mishchenko, M. Lozada, T. Georgiou, C. R. Woods, F. Withers, P. Blake, G. Eda, A. Wirsig, C. Hucho, K. Watanabe, T. Taniguchi, A. K. Geim, R. V. Gorbachev, *Nano Lett.* **2014**, *14*, 3270.
- [21] L. Britnell, R. V. Gorbachev, R. Jalil, B. D. Belle, F. Schedin, M. I. Katsnelson, L. Eaves, S. V. Morozov, A. S. Mayorov, N. M. R. Peres, A. H. Castro Neto, J. Leist, A. K. Geim, L. A. Ponomarenko, K. S. Novoselov, *Nano Lett.* **2012**, *12*, 1707.
- [22] A. Ranjan, S. J. O'Shea, A. Padovani, T. Su, P. La Torraca, Y. S. Ang, M. S. Munde, C. Zhang, X. Zhang, M. Bosman, N. Raghavan, K. L. Pey, *ACS Applied Electronic Materials* **2023**, *5*, 1262.
- [23] A. Ranjan, N. Raghavan, M. Holwill, K. Watanabe, T. Taniguchi, K. S. Novoselov, K. L. Pey, S. J. O'Shea, *ACS Applied Electronic Materials* **2021**, *3*, 3547.
- [24] Y. Hattori, T. Taniguchi, K. Watanabe, K. Nagashio, *ACS Nano* **2015**, *9*, 916.
- [25] H. G. Kim, H.-B.-R. Lee, *Chem. Mater.* **2017**, *29*, 3809.
- [26] X. Wang, T.-B. Zhang, W. Yang, H. Zhu, L. Chen, Q.-Q. Sun, D. W. Zhang, *Appl. Phys. Lett.* **2017**, *110*, 053110.
- [27] J. Wang, S. Li, X. Zou, J. Ho, L. Liao, X. Xiao, C. Jiang, W. Hu, J. Wang, J. Li, *Small* **2015**, *11*, 5932.
- [28] W. Li, J. Zhou, S. Cai, Z. Yu, J. Zhang, N. Fang, T. Li, Y. Wu, T. Chen, X. Xie, H. Ma, K. Yan, N. Dai, X. Wu, H. Zhao, Z. Wang, D. He, L. Pan, Y. Shi, P. Wang, W. Chen, K. Nagashio, X. Duan, X. Wang, *Nat. Electron.* **2019**, *2*, 563.
- [29] T. Li, T. Tu, Y. Sun, H. Fu, J. Yu, L. Xing, Z. Wang, H. Wang, R. Jia, J. Wu, C. Tan, Y. Liang, Y. Zhang, C. Zhang, Y. Dai, C. Qiu, M. Li, R. Huang, L. Jiao, K. Lai, B. Yan, P. Gao, H. Peng, *Nat. Electron.* **2020**, *3*, 473.
- [30] H. Zhu, X. Qin, L. Cheng, A. Azcatl, J. Kim, R. M. Wallace, *ACS Appl. Mater. Interfaces* **2016**, *8*, 19119.
- [31] M. Yamamoto, S. Dutta, S. Aikawa, S. Nakaharai, K. Wakabayashi, M. S. Fuhrer, K. Ueno, K. Tsukagoshi, *Nano Lett.* **2015**, *15*, 2067.
- [32] S. Lai, S. Byeon, S. K. Jang, J. Lee, B. H. Lee, J.-H. Park, Y.-H. Kim, S. Lee, *Nanoscale* **2018**, *10*, 18758.
- [33] K. Liu, B. Jin, W. Han, X. Chen, P. Gong, L. Huang, Y. Zhao, L. Li, S. Yang, X. Hu, J. Duan, L. Liu, F. Wang, F. Zhuge, T. Zhai, *Nat. Electron.* **2021**, *4*, 906.
- [34] K. Yang, T. Zhang, B. Wei, Y. Bai, S. Jia, G. Cao, R. Jiang, C. Zhang, E. Gao, X. Chang, J. Li, S. Li, D. Zhu, R. Tai, H. Zhou, J. Wang, M. Zeng, Z. Wang, L. Fu, *Nat. Commun.* **2020**, *11*, 2502.
- [35] M. Houssa, P. W. Mertens, M. M. Heyns, *Semicond. Sci. Technol.* **1999**, *14*, 892.
- [36] J. Sune, E. Y. Wu, W. L. Lai, *IEEE Trans. Electron Devices* **2004**, *51*, 1584.
- [37] R. Degraeve, T. Kauerauf, M. Cho, M. Zahid, L. A. Ragnarsson, D. P. Brunco, B. Kaczer, P. Roussel, S. D. Gendt, G. Groeseneken, in IEEE International Electron Devices Meeting, 2005, IEDM Technical Digest, **2005**, <https://ieeexplore.ieee.org/abstract/document/1609364>.
- [38] E. Simoen, B. Kaczer, M. Toledano-Luque, C. Claeys, *ECS Trans.* **2011**, *39*, 3.
- [39] A. Ranjan, N. Raghavan, K. Shubhakar, S. J. O'Shea, K. L. Pey, in *Noise in Nanoscale Semiconductor Devices* (Ed.: T. Grasser), Springer International Publishing, Switzerland, **2020**, pp. 417–440.
- [40] S. Vecchi, P. Pavan, F. M. Puglisi, *J. Appl. Phys.* **2023**, *133*, 114101.
- [41] A. Ranjan, F. M. Puglisi, N. Raghavan, S. J. O'Shea, K. Shubhakar, P. Pavan, A. Padovani, L. Larcher, K. L. Pey, *Appl. Phys. Lett.* **2018**, *112*, 133505.
- [42] A. Ranjan, F. M. Puglisi, J. Molina-Reyes, P. Pavan, S. J. O'Shea, N. Raghavan, K. L. Pey, *ACS Applied Electronic Materials* **2022**, *4*, 3909.
- [43] E. Y. Wu, R. P. Vollertsen, *IEEE Trans. Electron Devices* **2002**, *49*, 2131.
- [44] W. Weibull, *J. Appl. Mech.* **1951**, *18*, 293.
- [45] D. J. Dumin, *Oxide Reliability: A Summary Of Silicon Oxide Wearout, Breakdown, And Reliability*, 1<sup>st</sup> Edition, World Scientific Publishing Co Pte Ltd, Singapore, **2002**.
- [46] A. Ranjan, S. J. O'Shea, M. Bosman, J. Molina, N. Raghavan, K. L. Pey, in 2020 IEEE International Reliability Physics Symposium (IRPS), **2020**, <https://ieeexplore.ieee.org/abstract/document/9128353>.
- [47] C.-N. Huang, J.-S. Bow, Y. Zheng, S.-Y. Chen, N. Ho, P. Shen, *Nanoscale Res. Lett.* **2010**, *5*, 972.
- [48] E. Long, S. O'Brien, E. A. Lewis, E. Prestat, C. Downing, C. S. Cucinotta, S. Sanvito, S. J. Haigh, V. Nicolosi, *npj 2D Materials and Applications* **2017**, *1*, 22.
- [49] T. Zanotti, A. Ranjan, S. J. O'Shea, N. Raghavan, R. Thamankar, K. L. Pey, F. M. Puglisi, *IEEE Transactions on Device and Materials Reliability* **2024**, *24*, 184.
- [50] H. Yang, X. Chen, D. Wu, X. Fang, *Adv. Electron. Mater.* **2023**, *9*, 2300277.
- [51] S. Li, X. Liu, H. Yang, H. Zhu, X. Fang, *Nat. Electron.* **2024**, *7*, 216.
- [52] R. F. Egerton, *Electron Energy-Loss Spectroscopy in the Electron Microscope*, 3<sup>rd</sup> Edition, Springer, Berlin, **1996**.