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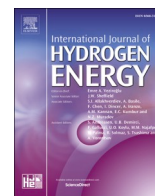
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Cofiring of hydrogen and pulverized coal in rotary kilns using one integrated burner

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ABSTRACT

The grate-kiln process for iron-ore pellet induration utilizes pulverized coal fired burners. In a developed infrastructure for H₂, it might be desirable to heat the existing rotary kilns with renewably produced H₂. Technical challenges of H₂ heating of grate-kilns include high emissions of NO_x and maintaining sufficient heat transfer to the pellet bed. This article examined cofiring (70% coal/30% H₂) in 130 kW experiments using two different integrated burner concepts. Compared to pure coal combustion, cofiring creates a more intense, smaller flame with earlier ignition and less fluctuations. The process temperature and heat transfer are enhanced in the beginning of the kiln. The co-fired flames emit 32% and 78% less NO_x emissions compared to pure coal and H₂ combustion, respectively. We can affect the combustion behavior and NO_x emissions by the burner design. H₂/coal cofiring using integrated burners is probably an attractive solution for emission minimization in rotary kilns.

1. Introduction

Rotary kilns are used to heat solid materials to high temperatures (>1000 °C) in a continuous process, for example in cement-, pulp and paper-, and iron ore industries [1–4]. The cement industry alone contributes to ~5% of global greenhouse gas (GHG) emissions whereas the iron and steel industry emits ~7% of global GHG emissions [5,6]. Roughly half of the total CO₂ emitted from the cement industry is from combustion of fossil fuels in the rotary kilns used for calcination and clinker formation; the remaining is emitted from the calcination of the limestone [5]. In the iron and steel industry, combustion in kilns is used in the sintering process of iron ore pellets inside the so-called grate-kiln (GK) induration process [7]. Consequently, rotary kiln operators must heavily reduce the fossil fuel usage in their furnaces to be able to reach the 1.5 °C global temperature increase target, signed by 196 parties in the 2015 Paris agreement [8] and the related targets by the European Union to reduce CO₂ emissions by 55% by 2030 (relative to 1990) and become climate neutral by 2050 [9].

According to Julian [10], carbon neutral iron-ore induration processes, for the iron and steel industry, should be developed promptly. Pelletizing plants will be in operation for a foreseeable future, therefore,

retrofitable solutions for reducing their emissions are essential. In Europe, focus is on electrification and hydrogen-based steelmaking [10]. In a developed infrastructure for H₂, it may be desirable to apply H₂ also for heating of the grate kiln induration process.

Replacing a fraction of the coal fuel in the kiln with H₂ gas from renewable powered water electrolysis will reduce the CO₂ emissions proportionally to the mixing ratio of the fuels. To our best knowledge no investigations of H₂ and coal cofiring in a single integrated burner sharing a single flame is reported in the literature. A challenge with H₂ combustion in grate kilns is high NO_x emissions [11,12]. Anthropogenic NO_x causes atmospheric ozone depletion, smog clouds, and acid rainfall [13]. Pure H₂ flames typically produce a larger amount of NO_x compared to fossil oils and coal due to the higher flame temperatures [14,15]. A high flame temperature allows the N₂ in the combustion air to form NO_x via the well-known Zeldovich mechanism [16]. By design, GK induration plants operate at large air-to-fuel equivalence ratio (λ), typically in the range of 4–6, for plants designed to oxidize magnetite (Fe₃O₄) to hematite (Fe₂O₃). The secondary combustion air in the process is preheated to temperatures above 1000 °C [17]. The large amount of excess secondary air required in the process along with the construction of the rotating kiln make it difficult to apply conventional

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primary NO_x mitigation strategies such as staged combustion [12]. Another potential obstacle with H₂ combustion in kilns is that the flame does not contain any radiating solid particles (i.e., soot, char and fly ash). It has previously been shown that radiation from the flame dominates the heat flux to the wall during coal combustion in the kiln [11, 18]. The potential change in radiation properties and shape of H₂ flames can potentially reduce heat transferred to the pellet bed affecting the production rate, quality of the product, and furnace efficiency negatively.

NO_x formation and heat transfer from combustion of solid and gaseous fuels have been studied extensively, see for example [16,19,20]. Furthermore, there are also studies of cofiring solid and gaseous fuels, where some of the investigations also involves H₂ [21–25]. However, all these studies applied a separate flame for each fuel, and with focus on other technical applications than rotary kilns. Pisa et al. [21] experimentally studied the influence of hydrogen enriched gas (HRG). HRG was premixed with primary air and pulverized brown coal in a burner installed in a 2 MW pilot furnace. They found that SO₂ emissions decreased while NO_x emission was increased with the addition of HRG in the primary air. Kim et al. [26] developed a CFD model of a 550 MW tangentially fired pulverized coal boiler co-fired with CH₄ in a second combustion stage. They found that introducing CH₄ up to 40% of total power in a secondary combustion zone decreased the predicted NO_x emissions by up to 70%. The decrease in NO_x pollution is explained by less production of fuel NO_x due to a lower flow of coal and by NO_x reburning in the CH₄ combustion zone. Since CH₄ contains carbon atoms, the reduction of CO₂ emissions was not severe. The amount of unburned fuel was reduced when the CH₄/coal ratio was increased.

There are several recent studies on co-firing of coal and NH₃ [27–32]. NH₃ is a H₂ rich chemical with promising properties as a renewable H₂ carrier that is combustible directly without requiring a H₂ extraction process [33]. Although NH₃ is H₂ rich, the physical properties of the chemical and its combustion behavior differs from pure H₂ combustion [29]. Aoyang et al. investigated co-firing of 0–50% NH₃ (on power basis) and pulverized coal in a drop tube furnace [27]. The authors found that co-firing with NH₃ increases the NO_x emissions and drastically reduces the SO₂ flue gas emissions compared to pulverized coal combustion. Co-firing of NH₃ leads to higher concentrations of submicron particles [27]. Tamura et al. studied co-firing of coal and NH₃ in a 1.2 MW bench scale burner [32]. Experiments were performed with a small amount of excess oxygen O₂ <5% with air staging technology to reduce NO_x emissions. The NH₃ was injected differently into the process, either premixed with coal, into the coal burner or separately at the side wall. Side wall injection increases the NO_x emissions while the NO_x emissions using premixed and burner gun injection remained at the same level as coal for mixing ratios between 0 and 30% NH₃ on energy basis. However, further increase of NH₃/coal blending ratio results in increased NO_x emissions. The amount of unburned carbon decrease also as the mixing ratio of NH₃ increased.

In this work, we study cofiring of 30% (energy basis) H₂ and non-premixed pulverized coal in a pilot scale furnace (130 kW) designed to simulate the combustion process inside rotary kilns [11]. The combustion characteristics of H₂/coal cofiring is compared to pure pulverized coal and pure H₂ combustion. Two burner concepts are investigated: i) an annular H₂ injection around a central coal injection and ii) an annular coal injection around a central H₂ injection. Moreover, several H₂ injection velocities are investigated in both burner concepts. The effect of H₂/coal cofiring on NO_x emissions and heat transfer relevant for rotary kiln applications are discussed.

2. Experimental conditions

2.1. Experimental setup

This study was performed in the test furnace called the horizontal industrial combustion kiln (HICK) located at RISE in Piteå, Sweden. The

HICK, which has a maximum fuel capacity of 150 kW, is a pilot-scale facility designed to simulate combustion conditions inside different types of rotary kilns. A sketch of the HICK with the system components and the location of the measurement ports are shown in Fig. 1. The HICK consists of six main parts: the fuel injection system, the burner hood, the kiln, the transfer chute, the boiler, and the flue gas channel.

Each fuel has a dedicated fuel feeding system. H₂ is supplied from a pressurized packages of 24 tubes (50 dm³ each at 200 bar pressure) located outside the building. The pressure in the gas transporting pipe was decreased by a pressure regulator to ~12 bar. The feeding rate of H₂ to the burner was regulated by a Bronkhorst (EL-FLOW Select F-203 A V) mass flow controller (MFC). The pulverized coal was supplied from a 0.6 m³ fuel hopper placed on top of weight cells for coal mass flow rate measurements. The coal feeding system dispatched the coal trough screw feeding into a fuel shaft for further transportation with transport air to the burner. The mass flow of coal was controlled by the rotation speed of the screws. Transport air was fed (through an ejector) to the fuel shaft by an MFC (Red-y, GSC-D5SS-BB13). Burner air was fed to the burner separately through a pipe controlled with a separate MFC (Brooks, SLA5800 Series).

The burner hood consists of a burner unit and a preheating unit (denoted 3 and 4 in Fig. 1). In commercial induration plants, the combustion air is preheated by waste heat to increase the efficiency of the system [34]. To mimic this, the HICK has a preheating system consisting of eight 15 kW air heaters (Leister, LE 10 000 H T) with a maximum capacity of 20 dm³/s that preheats the air to ~850 °C based on S-type fine wire thermocouple measurements. The preheated secondary air is separated vertically in two rows by a horizontal plate containing the burner.

Two conceptual burner designs with several configurations each were used, see the nozzles sketched in Fig. 2 and drawings of the burners are presented in Fig. S1 in the supplementary material. In burners denoted O-, coal is feed through the central pipe and H₂ through an annular pipe with a plate with 1–8 holes with a diameter of 2 mm. In burners denoted I-, H₂ was injected through the central pipe through an opening with a diameter of 2–6 mm, while coal was feed through the annular pipe. A primary air register without swirl provides combustion air for ignition and cooling of the burner tip in both burner concepts. The burner tip is inserted 150 mm into the kiln section (see x-axis in Fig. 1).

The brick lined horizontal furnace has a length of 3300 mm and has a square shaped cross section with an inner height and width of 550 mm. The process temperature (T_{proc}) is measured with six thermocouples (type K, 3 mm) located along the roof of the kiln. The thermocouples barely penetrate the insulation, a few millimeters into the gas stream. The kiln has several access ports for measurements. Most relevant to this study are three optical ports (P2, P4, and P5) and two ordinary measurement ports suitable for probe measurements (P1 and P3), see Fig. 1. P2 and P4 are located on the front side of the kiln at 970 mm and 2500 mm from the start of the kiln respectively. P5 is located at the back of the kiln, opposite the front of the burner and the hood. P1 and P3 are also on the front side of the kiln at 720 mm and 2000 mm from the beginning of the kiln.

The kiln gradually reduces at the end into the smaller 230 mm square shaped transfer chute that connects the kiln to the boiler. The T_{proc} of the transfer chute was measured with a 1.5 mm K-type thermocouple located in the center line of the transfer chute.

In the boiler, the flue gas is cooled down to ~400 °C. The HICK was operated at a sub-atmospheric pressure of ~15 Pa regulated by a flue gas fan in the exhaust gas system. There are small ports in the beginning of the chimney, located approximately 500 mm after the boiler outlet. These ports provide access for gas measurements of the flue gas composition.

2.2. Measurement equipment

The wet flue gas composition in the chimney was continuously

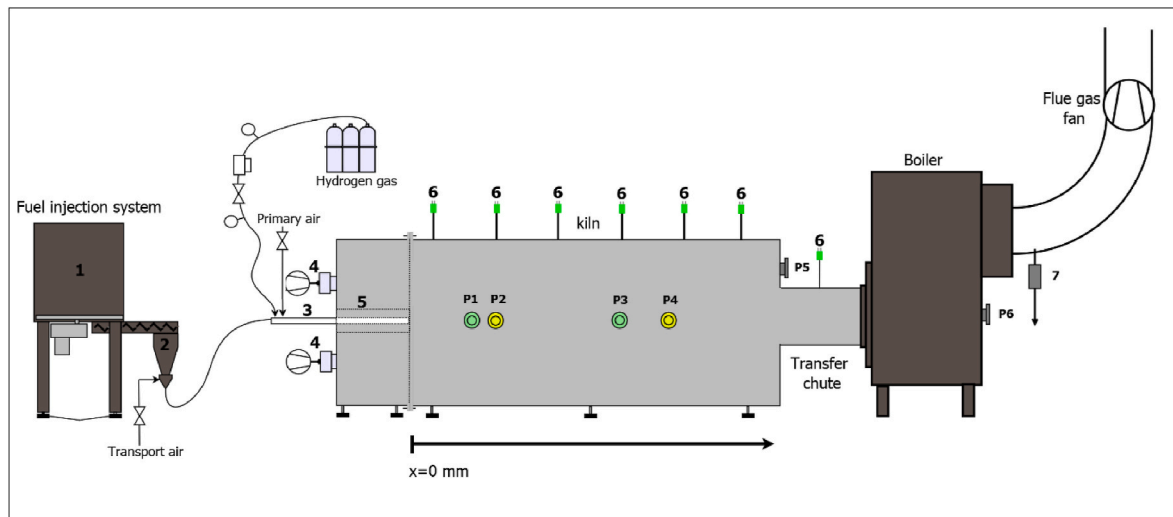


Fig. 1. The HICK and the experimental setup; (1) Fuel silo; (2) Fuel transport shaft; (3) Burner; (4) Preheated secondary air fans; (5) Burner hood; (6) Thermocouples; (7) Filtered flue gas access. The different measurement port P1–P6 are also shown in the Figure. P1 and P3 are used for heat transfer measurements. P2 and P4 are used for Tunable diode laser absorption measurements and emission measurements. The flame is video recorded through P5 and P6.

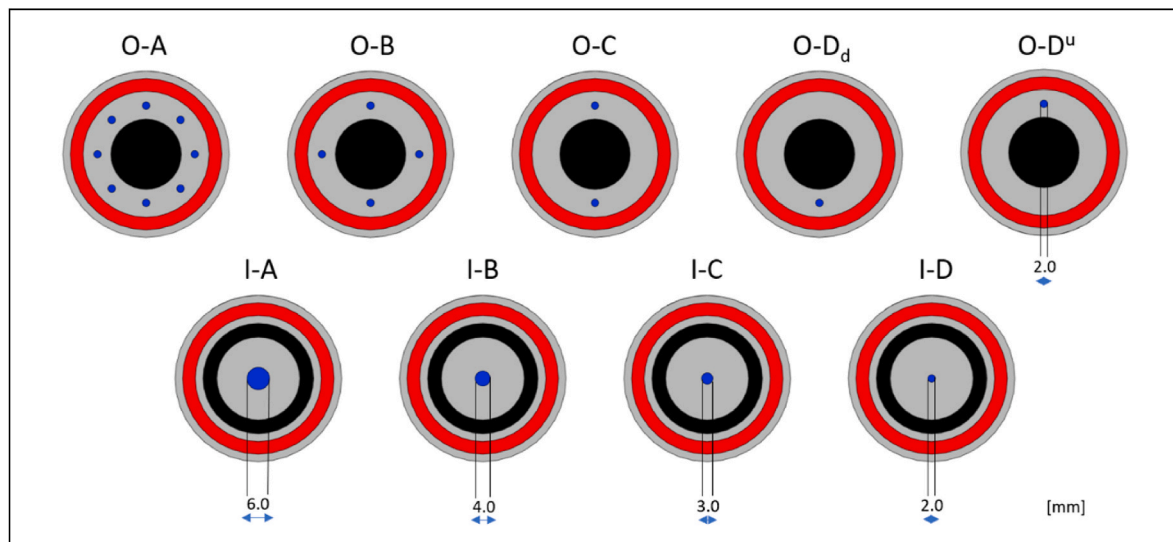


Fig. 2. Sketch of tested burner types, viewed from inside the kiln. Red areas represent primary air slit, the blue areas represent H_2 gas injection, and the black areas represents the coal injection. In burners O-, coal is injected centrally surrounded by H_2 injection holes and an annular primary air slit. In burners I-, H_2 is injected centrally with coal and primary air in annular slits surrounding the hydrogen injection. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

measured with respect to H_2O , CO_2 , CO , NO and NO_2 with a Fourier transform infrared spectrometer (MKS Multigas 2030 FTIR). In addition, the dry flue gas composition with respect to O_2 , CO_2 , CO , NO and NO_2 was measured with a flue gas analyzer using NDIR (non-dispersive infrared spectroscopy) in combination with a paramagnetic cell for O_2 with a traditional gas instrument (MRU Air MGA Prime). A heated filter filled with glass wool was used to remove particles in the gas drawn to the FTIR and MGA Prime. The gas was pumped to the instruments through a heated sampling line to prevent condensation, see (7) in Fig. 1. The gas composition for all combustion species except O_2 was determined using FTIR. The presented O_2 was measured with MGA Prime and then adjusted to wet basis using the FTIR-measured H_2O concentration. The heat transfer to the furnace wall was measured using an IFRF total heat flux probe that capture the total heat flux to a metal surface. Measurements were performed in ports P1 and P3 (see Fig. 1). The probe tip was aligned with the inner wall of the furnace and the

measurements were performed for 15 min or until a stable value was attained. Detailed information on the total heat flux radiometer can be found in Ref. [35].

Videos of the flames were filmed through port 5 using a Samsung S22 S901B smartphone at a 1080×1920 resolution. No manual editing of the footage was performed except for cropping and trimming. Tunable diode laser absorption spectroscopy (TDLAS) and emission measurements were performed through ports P2 and P4 in Fig. 1.

2.3. Spectroscopic measurements

The gas temperature (T_{gas}) and H_2O concentration were measured using a TDLAS system which consists of a Nanoplus distributed feedback laser diode generating light at approximately $1.4 \mu m$, a controller (Thorlabs, ITC4001), a photodetector (Thorlabs, PDA10) and a data acquisition system (National Instruments, with NI PXIe-6356 card). A

more detailed description of the laser system, the procedure for data treatment, and sensor application examples can be found elsewhere [36–38]. The overall uncertainties in HICK TDLAS measurements were 75 °C and 15% (relative) for temperature and H₂O concentration. Emission spectra were measured using an Ocean Optics spectrometer (HR2000 + GC, HC-1 Grating, 600 gr/mm, blazed @ 300 nm) in the wavelength range from 200 nm to 1100 nm. The spectrometer was intensity-calibrated using a blackbody calibration source (Dias, PYROTHERM CS 1500). The radiation was collected by collimating lens ($f = 8$ mm) which focused the radiation into an optical fiber (with 400 μm core diameter) coupled with the spectrometer. The temperature of the particles in the flame (T_{particle}) was estimated by fitting the recorded spectra with a Planck function, for a more detailed description of the procedure see Ref. [11]. When the dominating radiation was the furnace wall, the wall temperature (T_{wall}) was determined.

2.4. Fuel analysis

Coal powder and H₂ were used as fuel in this study. The ultimate and proximate analysis as well as the particle size distribution of the coal can be seen in Table 1. The analysis also contains the relative uncertainties and references to the methods used to acquire the fuel analysis. The H₂ used in this study had a purity > 99.9 vol-%.

2.5. Experimental conditions

Preheating started with a 100-kW side mounted diesel burner installed in port P1 for about 18 h, followed by 2 h of heating with a 175 kW centrally mounted diesel burner. During central heating, the air preheating system was engaged to achieve a temperature profile in the furnace similar to the experimental kiln conditions. Finally, about an hour of pure coal combustion was performed to further heat the kiln to a stable temperature profile. After preheating the kiln, the integrated H₂/coal burner was installed. Between testing of different burners, a sequence of pure coal combustion was performed to avoid heat loss. Each burner configuration was tested until stable readings on the gas measurement instruments had been attained and heat transfer measurements had been performed, taking about 40–60 min per

Table 1
Coal fuel analysis.

	Unit	Measure	Analysis method
Particle size distribution			
<1 μm	%	2.1	Sieve
<5 μm	%	15.6	Sieve
<10 μm	%	29.2	Sieve
<15 μm	%	39.5	Sieve
<20 μm	%	47.6	Sieve
<45 μm	%	71.7	Sieve
<63 μm	%	81.3	Sieve
<90 μm	%	89.8	Sieve
<125 μm	%	95.2	Sieve
<180 μm	%	98.5	Sieve
<250 μm	%	99.6	Sieve
Proximate analysis (as received)			
Moisture	%	0.5 \pm 0.05	ISO 589:2018 mod
Volatile matter	%	19.2 \pm 1	ISO 562:2010 mod
Fixed carbon ^a	%	68.2 \pm 5	ISO 562:2010 mod ^a
Ash	%	12.1 \pm 0.6	ISO 1171:2018
Ultimate analysis (as received)			
Cl	%	<0.01	ASTM D4208-2019 mod
S	%	0.3	ASTM D4239-2018
C	%	77.0	ASTM-D5373:2016
H	%	3.9	ASTM-D5373:2016
N	%	1.33	ASTM-D5373:2016
O ^a	%	5.3	ASTM D3176-2015 ^a
Calometric analysis (as received)			
LHV	MJ/kg	30.473	ISO 1928:2020

^a Denotes calculated values.

configuration. During cofiring the kiln was operated at 130 ± 1 kW with H₂ accounting for 32% of the power (~ 41 kW) and coal for the remaining 68% (~ 89 kW). Combustion was performed at a stoichiometric ratio of approximately $\lambda = 3.6$, that is with an air flow of 6.66 nl/s going through the burner primary air register and a sum of 6400 dm³/s through the eight electric air preheaters. The primary air used to transport the coal was 2.50 dm³/s when cofiring and 3.33 dm³/s for coal combustion. The operational conditions during the experiments are detailed in Table 2. The coal feeding rate is calculated from fuel silo weight measurements during operation. The actual secondary air flow rate and the resulting stoichiometric ratio is calculated from the gas composition measured in the chimney (7 in Fig. 1).

3. Results and discussions

3.1. Flame characteristics

Video recordings of the flames (supplementary video 1 and video 2) illustrates the differences in flame characteristics of H₂/coal cofiring compared to pure coal and pure H₂ combustion. Representative snapshots of the flames extracted from the videos are presented in Figs. 3 and 4. We strongly encourage the reader to watch the videos in the supplementary material for an illustration of the dynamic combustion process. In general, the flames are centrally located in the furnace. Independent of burner configuration, the coal flame (Fig. 3a and 4a and video 1 and 2) is less bright and noticeably less compact compared to the H₂/coal cofiring flames (Fig. 3c and 4c and video 1 and 2). On short time scales (<1 s) the coal flame flickers due to turbulence. On larger time scales (>1 s) the overall intensity and point of ignition varied significantly resulting in a back and forward pulsating flame (see Video 1 and 2). This is due a variation in the fuel feeding rate, as observed previously during coal and biomass combustion in the HICK [11]. The pure H₂ flame is visible as a blue flame in the central part of the furnace (Fig. 3b and 4b). Recent findings suggest the blue emission comes from the radiative recombination of H + OH and OH + OH [39]. As seen here, the blue emission is so strong that it makes the H₂ flame visible although there is other black body radiation from the hot walls of the furnace that interferes in the visible spectrum of the light.

Compared to pure coal combustion, the H₂/coal cofiring resulted in a more compact and brighter flame that is more constant in flame size and with a higher flickering frequency (see Fig. 3c and 4c), probably due to higher turbulence combustion intensity from the H₂ jets in the burner. At the same time the pulsation of the flame is reduced. This observation suggests an earlier and more consistent ignition of the coal particles in the H₂/coal flame. Probably, the heat generated from H₂ combustion by the easily ignited H₂ injected into the coal particle stream serves both as a heat source for devolatilization of the coal particles and stable ignition source of the devolatilized gas. This has been previously observed for co-combustion of natural gas and coal powder flames [40]. As a result, the ignition and combustion of coal particles occurs closer to the burner, and pulsation is reduced.

Comparing the cofired burner designs (Fig. 3c–g and 4c to 4f), the burners with higher H₂ injection velocity give a high frequency flickering and brighter flame with a lower cross-sectional area covering less of the view from the porthole. This indicates that higher H₂ injection velocities produce compacter (in radial direction) and more intense flame at the centerline of the furnace. This could be a result of an increased entrainment of the preheated surrounding air into the flame, caused by the higher momentum of the H₂ jets in the burner which in turn increase the turbulence intensity and mixing of air and fuel in the jet flames, generating higher combustion intensity along the center line of the furnace.

The time resolved TDLAS measurements from port P2 and P4 for all burner cases are presented in Fig. 5 for T_{gas} and Fig. 6 for H₂O below. Tabulated average data and standard deviations from the TDLAS measurements are available in Tables S1 and S2. During coal combustion

Table 2
Operational conditions during experiments.

Burner	H ₂		Coal		H ₂ + Coal								
	O-A	I-A	O-A	I-A	O-A	O-B	O-C	O-D _d	O-D ^{II}	I-A	I-B	I-C	I-D
H ₂ Velocity [m/s]	517	460	–	–	155	310	620	1241	1241	138	310	552	1241
Power [kW]	139 ± 0		128 ± 2		130 ± 1								
Power H ₂ [kW]	139		0		41								
Power coal [kW]	0		128 ± 2		89 ± 1								
Feeding rate coal [kg/h]	0		15.1 ± 0.2		10.5 ± 0.1								
Transport air flow rate [dm ³ /s]	0		3.33		2.50								
Transport air temp [°C]	–		20		20								
Transport air velocity [m/s]	0		11.7		8.8								
Primary air flow rate [dm ³ /s]	6.67		6.67		6.67								
Primary air temp [°C]	20		20		20								
Preheated Secondary air [dm ³ /s]	106 ± 2		106 ± 2		106 ± 2								
Lambda	3.59 ± 0		3.56 ± 0.05		3.60 ± 0.03								

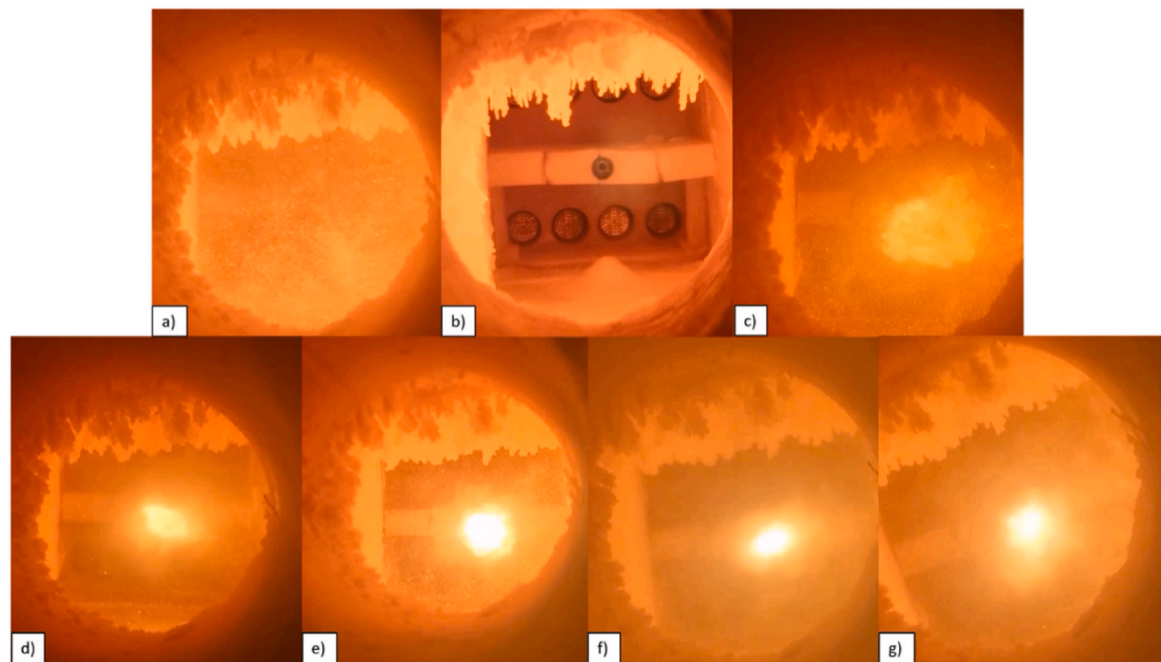


Fig. 3. Snapshots of the inside of the furnace when firing with hydrogen in the outer register. Coal combustion with burner O-A is shown in (a), H₂ combustion with burner O-A is shown (b), and co-fired combustion are shown in (c–g) with burner O-A in (c), burner O-B in (d), burner O-C in (e), burner O-D_d in (f) and burner O-D^{II} in (g).

there was a ~ 350 °C increase in T_{gas} (Fig. 5a and b) between port P2 and port P4 as well as an $\sim 100\%$ increase in H₂O concentration on average (Fig. 6a and b), indicating that a significant amount of the coal combustion occurs downstream of port P2. There are also large fluctuations in T_{gas} and the H₂O concentration in both P2 and P4 measurement positions (standard deviations 63 °C, 48 °C, 0.2% and 0.3%, respectively) indicates fluctuating combustion, which is in line with the flame observations. We did not evaluate the TDLAS measurements performed at P2 during pure H₂ combustions, since we expect temperatures well above 2000 °C at the center of the jet as well as sharp gradients of both T_{gas} and H₂O along the line of measurement, which makes the values unrepresentative of path-averaged temperatures and concentrations [41]. Therefore, only results from port 4 are presented in the figures for H₂ flames. Compared to pure coal combustion, the pure H₂ combustion conditions show almost no variations indicating almost constant combustion conditions and a stable flame. The average standard deviations of the temperature and H₂O concentration during pure H₂ combustion at P4 were 16 °C and 0.1%. Note that the average water concentration was 14.6 vol-% during pure H₂ combustion.

During H₂/coal cofiring, the average T_{gas} was on average 360 °C

higher at P2 but slightly (50 °C) lower at P4 (Fig. 5e and f) relative to pure coal combustion, which suggests earlier ignition and a shorter flame for the cofiring cases. Furthermore, the fluctuations in T_{gas} and H₂O concentration are significantly lower than during coal combustion at both ports, although not as low as during H₂ combustion (at P4), which supports the previous statement that H₂ injection stabilizes the coal combustion. The average standard deviation of the T_{gas} and H₂O measurements for the cofiring configurations at P2 is 23 °C and 0.2% and for P4, 25 °C and 0.25%. Considering the elevated H₂O concentration in the co-fired flames, both the normalized T_{gas} and H₂O standard deviations indicate a more stable combustion during co-combustion with H₂.

There is a minor effect of burner design concept on T_{gas} and H₂O concentration. At low H₂ injection velocities (140–310 m/s) the T_{gas} close to the burner is higher than in the middle section of the furnace, with an average T_{gas} of 1416 °C measured at port 2 compared to 1312 °C at port 4, see Fig. 5e and h. While T_{gas} decreases between ports, the H₂O concentration remains constant at 5.8%, see Fig. 6e and h, which indicates almost complete combustion already at Port 2 for low velocity burners. At high H₂ injection velocities (550–1200 m/s), the T_{gas} and

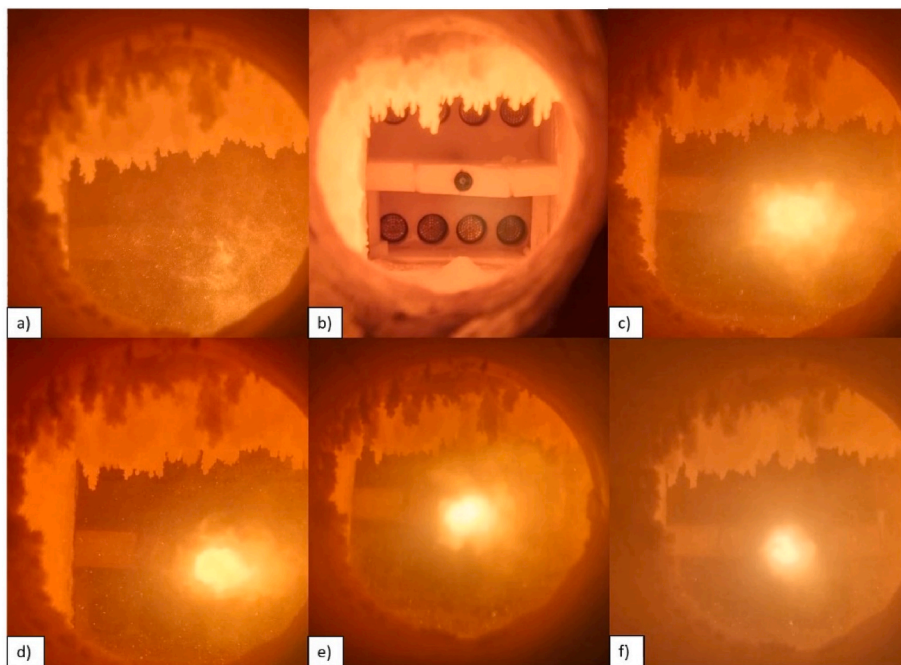


Fig. 4. Snapshots of the inside of the furnace when firing with hydrogen in the inner register. Coal combustion with burner I-A is shown in (a), H₂ combustion with burner I-A is shown in (b), and co-fired combustion are shown in (c–e) with burner I-A in (c), burner I-B in (d), burner I-C in (e), and burner I-D in (f).

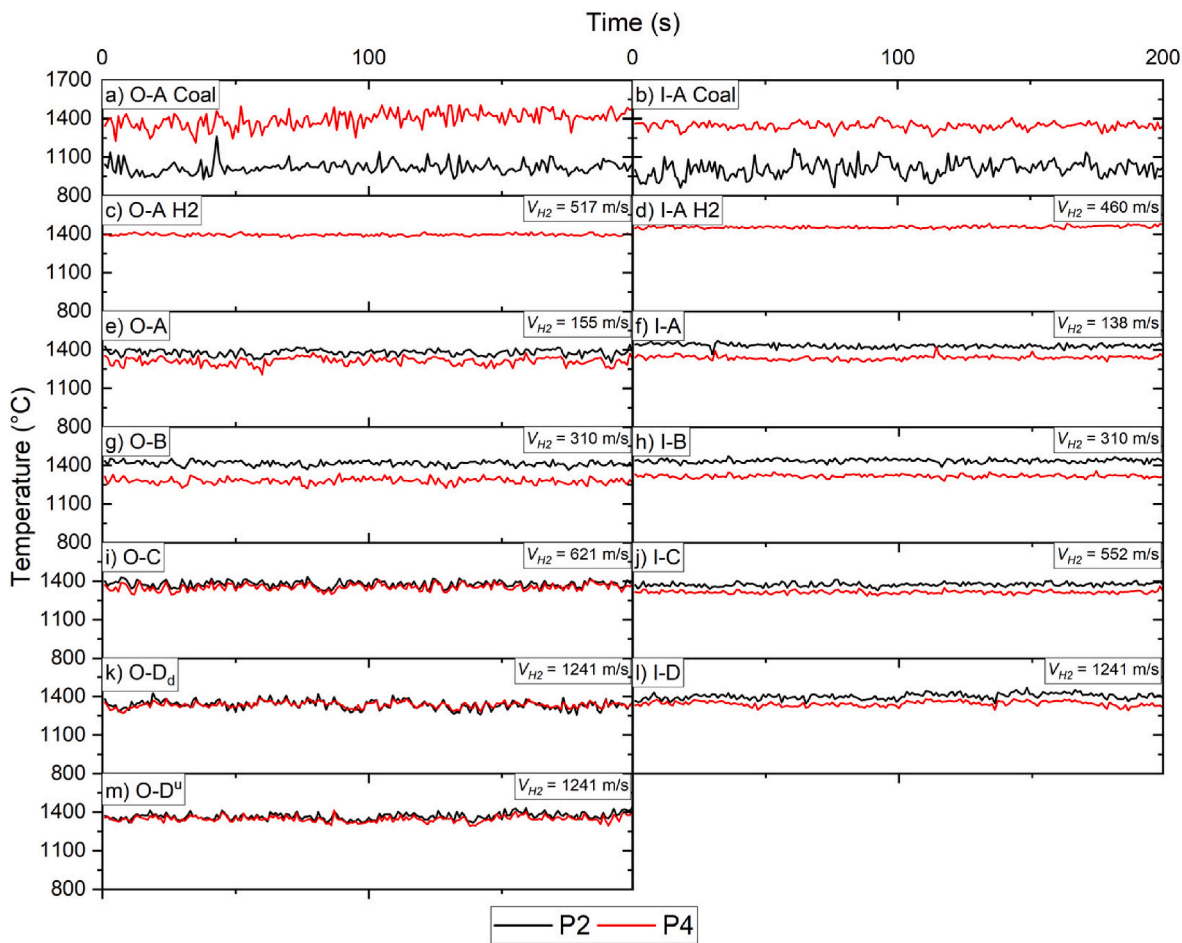


Fig. 5. Time-resolved TDLAS measurements of temperature at P2 and P4. (a) Coal combustion using the O burner. (b) Coal combustion using the I burner. (c) H₂ combustion using the O burner. (d) H₂ combustion using the I burner. (e), (g), (i), (k) and (m) corresponds to coal/H₂ cofiring with the O burner at different H₂ injection velocities. (f), (h), (j) and (l) corresponds to cofiring with the I burner at different H₂ injection velocities.

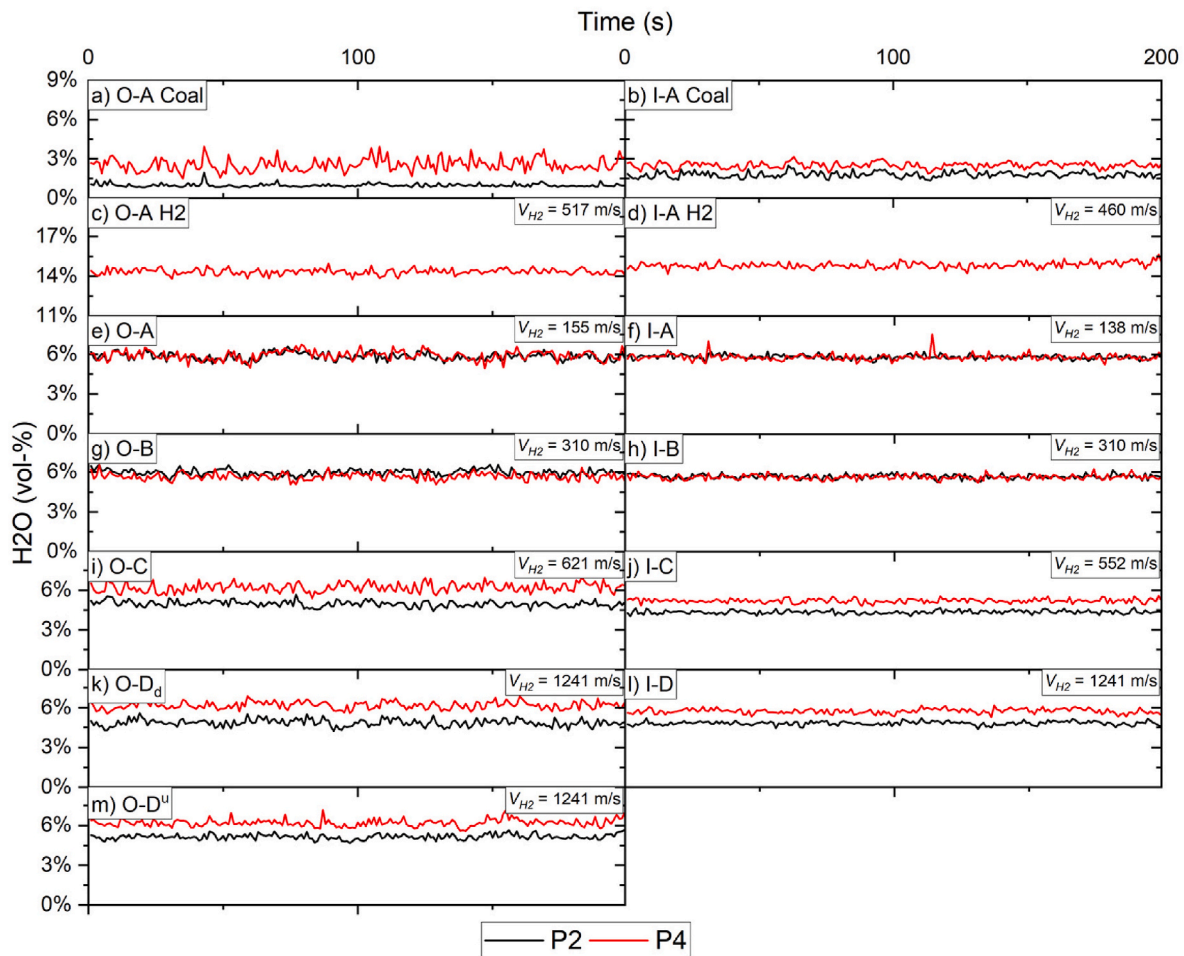


Fig. 6. Time-resolved TDLAS measurements of H₂O at P2 and P4. (a) Coal combustion using the O burner. (b) Coal combustion using the I burner. (c) H₂ combustion using the O burner. (d) H₂ combustion using the I burner. (e), (g), (i), (k) and (m) corresponds to coal/H₂ cofiring with the O burner at different H₂ injection velocities. (f), (h), (j) and (l) corresponds to cofiring with the I burner at different H₂ injection velocities.

H₂O concentration at Port 2, 1371 °C and 4.8% respectively, is decreased compared to the low velocity counterparts. At the same time, the T_{gas} at Port 4 is increased to 1339 °C with a similar H₂O concentration of 5.9% relative to the low velocity burners. The results indicate that combustion largely takes place upstream of port 2 for the low velocity burners, as the H₂O concentration remains constant between ports and there is a large decrease in T_{gas}. For the high velocity burners on the other hand, the results indicate that the flame is shifted forward by the increased momentum of the injected H₂, with combustion still ongoing between ports as suggested by the increasing H₂O concentration and a smaller drop in temperature between ports relative to the low velocity burners.

3.2. Process temperature and heat transfer

The averaged T_{proc}, T_{gas} and T_{particle} or T_{wall} at the ports P2 and P4 are shown in Fig. 7. During coal combustion, the temperature in the beginning of the furnace (T_{proc}) is ~100 °C lower compared to cofiring and ~80 °C lower compared to H₂ combustion, which indicates an earlier ignition with H₂ present. Therefore, the T_{proc} profile is relatively even in the furnace for the cofiring cases. The T_{proc} at the outlet of the kiln is similar for all cases, independent of the fuel.

Due to the pulsating flame, it is possible to measure both T_{wall} and T_{particle} at Port 2 by the emission spectroscopy. At P4, no (or low compared to wall radiation) coal particles are seen and only T_{wall} is measured. For all investigated cases, T_{wall} measured by the spectrometer

is in good agreement with corresponding T_{proc} measured with the thermocouples. The largest difference between T_{particle} and T_{gas} was measured for the pure coal flame, which is another indicator of late and varying point of ignition for this case. For the O burner concept, especially burner O-A and O-B, the T_{particle} is 150 °C lower than that for coal combustion (1700 and 1850 °C, respectively). A reasonable explanation for this observation is that H₂O produced from the H₂ combustion interact with the combustion of the coal particle to form an oxygen deficient zone surrounding the coal jet, which reduces the maximum T_{particle}. Since the H₂ is injected centrally and inside the flow of the coal powder, a similar reduction could not be observed for the I burner concept.

The averaged T_{gas} increases with 350 °C between P2 and P4 for coal combustion whereas during cofiring the T_{gas} is similar or even slightly reduced between the ports - again indicating earlier ignition during cofiring. Cofiring leads to a lower T_{gas} at P4 compared to coal combustion, this can partially be explained by elevated heat losses during cofiring where the measured heat transfer is larger in port 1. The difference in ignition between pure coal and cofiring can also change the vertical location of the flame at P4 due to buoyancy effects (lifted flame flames).

The results from the total heat flux radiometer is presented in Fig. 8 and available in tabulated form in Table S5. For all cases, the total heat flux were higher at P3, compared to those at P1. During coal combustion the total heat increases from 254 kW/m² to 374 kW/m² between P1 and P3. The increase in heat transfer to the wall between P1 and P3 is not as

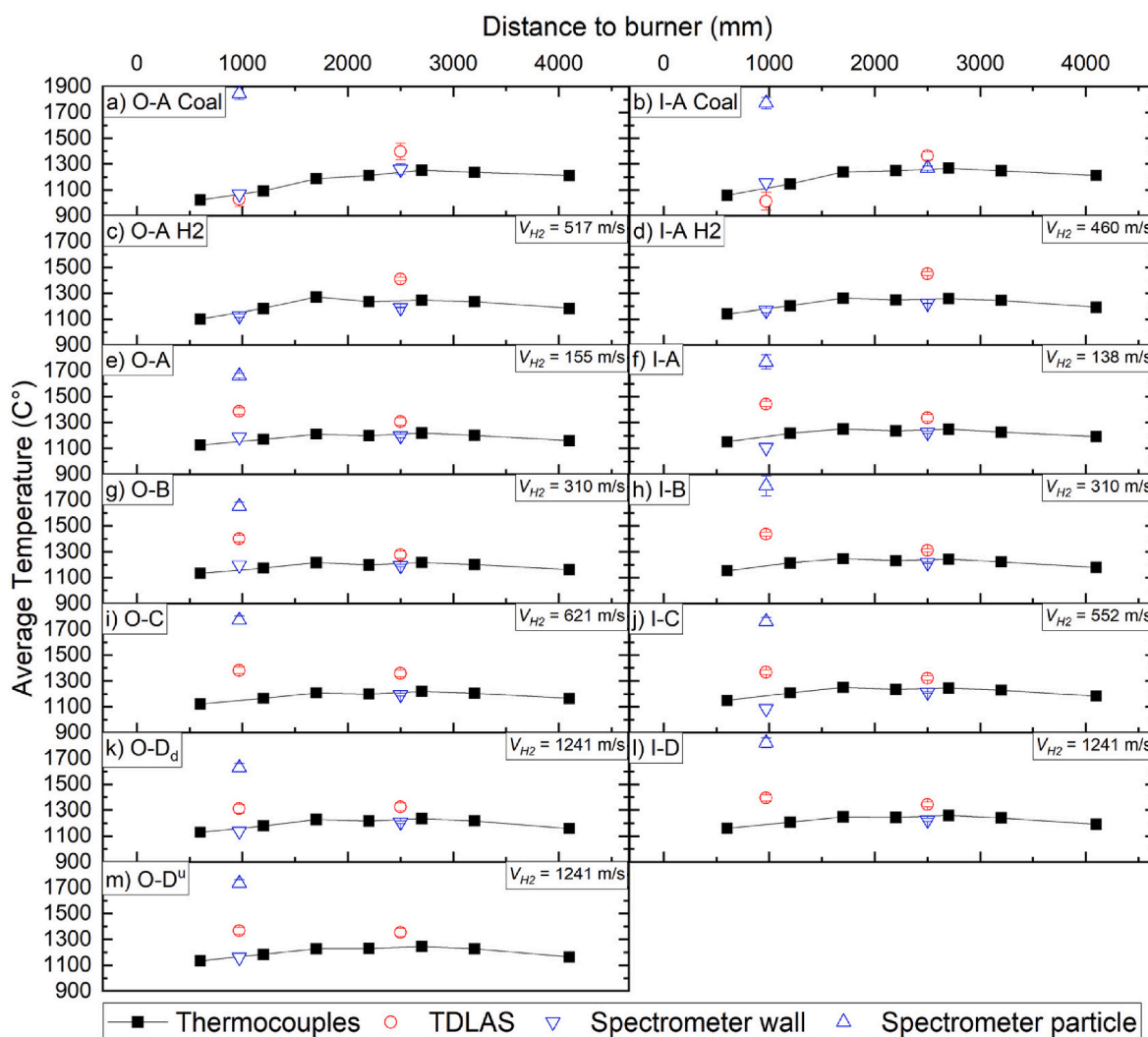


Fig. 7. Time-averaged temperature profiles measured in the furnace with the 7 thermocouples inserted through the roof of the kiln as well as the average TDLAS and spectrometer measurements in ports P2 and P4. (a) Coal combustion using the O burner. (b) Coal combustion using the I burner. (c) H₂ combustion using the O burner. (d) H₂ combustion using the I burner. (e), (g), (i), (k) and (m) corresponds to coal/H₂ cofiring with the O burner at different H₂ injection velocities. (f), (h), (j) and (l) corresponds to cofiring with the I burner at different H₂ injection velocities. Tabulated data is available in Table S3.

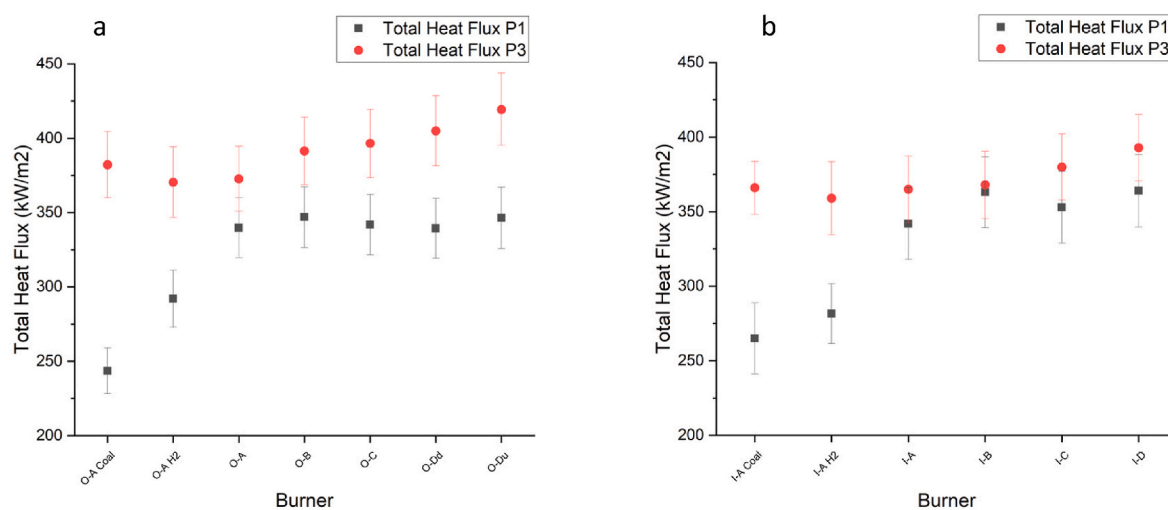


Fig. 8. Total heat flux at port P1 and port P3 measured using the total flux radiometer. (a) Corresponds to measurements using burner O and (b) corresponds to measurements using burner I.

prominent for pure H₂ and H₂/coal cofiring cases. Again, this behavior in heat transfer supports the discussion of delayed ignition and heat release for pure coal combustion compared to H₂ combustion and H₂/coal cofiring. Furthermore, cofiring results in larger total heat flux at P1 (348 kW/m²) compared to the total heat flux (254 kW/m²) of not only pure coal combustion, but also to pure H₂ combustion (287 kW/m²). At P3, the total heat flux for all fuels tested are on average on similar levels. The findings in pure H₂ and pure coal combustion support recent numerical simulations by Ehlme et al. which show that H₂ flames result in higher heat transfer rates to the kiln walls close to the burner compared to coal flames [42].

The total heat flux at P3 was found to vary slightly with burner configuration when cofiring H₂ and coal. Increasing H₂ velocity results in higher total heat flux in the middle section of the furnace (P3). This is likely due to the increased turbulence intensity induced by the higher velocity H₂ jets which in turn increases the turbulence levels in the kiln downstream.

3.3. Emissions

Fig. 9 shows the measured flue gas composition (wet basis) measured at the outlet (see 7 in Fig. 1). The wet gas compositions from the FTIR and the MGA Prime instruments are presented in Table S4. The major gas species (O₂, H₂O, and CO₂) as well as CO show similar levels during repetitions. Furthermore, the low CO levels (<15 ppm) throughout all experiments indicate near complete combustion. The FTIR (exhaust) and TDLAS (P4) H₂O data are within the measurement accuracy for all experimental cases.

An average outlet O₂ concentration of 14.6% was measured in all cases except during H₂ combustion where the average O₂ concentration was 13.7%. The lower oxygen concentration during H₂ firing is explained by a slightly higher power: 139 kW instead of 130 kW as during coal and cofiring; note that air flow was kept constant throughout all experiments. The coal combustion resulted in 5.6% CO₂ in the flue gas on average. The carbon-free H₂ naturally resulted in zero CO₂ and cofiring resulted in an average CO₂ concentration of 3.9%. Replacing 30% of the coal during H₂/coal cofiring resulted in 31% decrease in outlet CO₂.

The NO_x emissions are presented in Fig. 10 as the averaged NO₂ mass emitted per unit of energy (mg/MJ_{Fuel}), calculated from the measured outlet NO_x concentration and the flue gas flow rate estimated by assuming complete combustion. Cofiring was found to reduce NO_x emissions compared to coal combustion, with both burner concepts (O and I) resulting in similar NO_x emission levels. The average NO_x emission for the coal combustion cases was 915 mg NO₂/MJ, 2871 mg NO₂/MJ for the H₂ combustion cases, and 593 mg NO₂/MJ for the H₂/coal cofiring cases. Pure H₂ combustion produces most NO_x, due to the Zeldovich reactions associated with a high flame temperature. Cofiring H₂ and coal lead to a reduction of the NO_x formation of 35% on average compared to pure coal and a reduction of 79% compared to pure H₂ combustion. The decrease of NO_x during cofiring suggests that the (maximal) flame temperature is no longer sufficiently high to activate the substantial NO_x production due to Zeldovich mechanism once H₂ is introduced to the coal flame. This is possible provided that a significant part of the early heat release of the H₂ combustion is transferred to the coal particles. This explanation is consistent with the earlier ignition of the coal particles. The larger reduction in NO_x emission than the reduction in the mass influx of fuel-bound N in the low velocity H₂ cofiring burners O-A, O-B and I-C indicates the possible importance of NO reburning or char reduction pathways under the conditions studied. For example, it is well known in coal combustion that higher devolatilization temperatures yields more N release (mostly as HCN and NH₃) in the devolatilization stage [17], which can reduce NO through reactions forming N₂. The argument of early heat release of H₂ igniting the coal would also increase the devolatilization temperature of the coal particles. We plan to investigate the NO_x formation in the system during

cofiring conditions numerically in our future work.

The design of the cofiring burner affected the NO_x formation by about 10%. Burners of type O generated on average 567 mg NO₂/MJ of NO_x while burners of type I generated on average 601 mg NO₂/MJ. The only burner of type O that produced more NO_x than its counterpart of type I was burner O/I-D, with the highest velocity, which had a 16 mg NO₂/MJ larger formation.¹ For the other burners, H₂ injected through the outer register generated on average 51 mg/MJ less NO_x compared to their counterparts with H₂ injected through the central register. No significant difference in NO_x formation was found between the two configurations of burner O-D, O-D_d and O-D^u. This observation is not intuitively obvious, as H₂ combustion has been shown to yield lifting flames [11], meaning that for burner O-D_u the hydrogen flame would lift away from the coal flame while for burner O-D_d it would instead lift into the coal flame. This independence of orientation could be an indication that with these burner concepts the H₂ and coal mixes well and the mixing is robust still producing a single flame.

Fig. 11 shows the NO concentration as a function of H₂ injection velocity. The NO_x formation increases almost linearly with H₂ velocity. Increased inlet velocities likely generate higher turbulence intensity in the flame. A more intense flame is known to favor NO_x formation as it reduces the residence time of the N-bound intermediates inside the reducing environment. The detailed effect of inlet velocity of the H₂ stream will, however, require further investigations.

4. Conclusions

In this work, co-combustion of H₂ and pulverized coal has been studied in a 150-kW furnace developed to simulate the conditions of the kiln in an iron ore grate-kiln induration machine. The fuels were fed through separate registers integrated in the same burner.

The results show that replacing 30% (energy) of the coal with H₂ results in a flame that is visually smaller, brighter and with less fluctuations regardless of burner design. The addition of H₂ stabilizes and enhances the ignition of coal particles and the combustion process. Cofiring of H₂ and coal generates the higher wall temperatures in the early part of the kiln due to an increased heat transfer compared to both pure H₂ and pure coal firing. The heat transfer in later parts of the kiln is similar for all fuels. Furthermore, the results showed that the H₂/coal cofiring generated 32% less NO_x emissions (623 mg/MJ) on average compared to coal combustion (915 mg/MJ) and 78% less than H₂ combustion (2870 mg/MJ). The decreased NO_x formation during cofiring is explained by a combination of early coal ignition, decreased amount of fuel bound nitrogen, and possibly suppression of the high flame temperatures typically found in H₂ flames due to heat transfer to the coal particles. The results indicate that the H₂ injection velocity has an impact on the NO_x emissions, with less NO_x emissions at lower H₂ velocities. The lowest H₂ injection velocity tested for each burner concept, i.e. burners O-A and I-A (155 and 138 m/s) had the lowest NO_x emissions (520 and 557 mg/MJ). The lowest NO_x emissions, the low intensity flame (relative to the co-fired flames), and a more conventional coal injection design through a O-burner makes the O-A burner the most suitable for practical application. Overall, our findings support that replacing a portion of the coal with H₂ using one integrated burner in existing coal fired induration fueled rotary kilns is an attractive solution to reduce CO₂ and NO_x emissions.

CRedit authorship contribution statement

Andreas Johansson: Writing – review & editing, Writing – original draft, Visualization, Validation, Investigation, Formal analysis, Data curation, Conceptualization. **Johannes Fernberg:** Writing – review &

¹ Calculated by considering the NO_x level of burner O-D as the average of the two orientations.

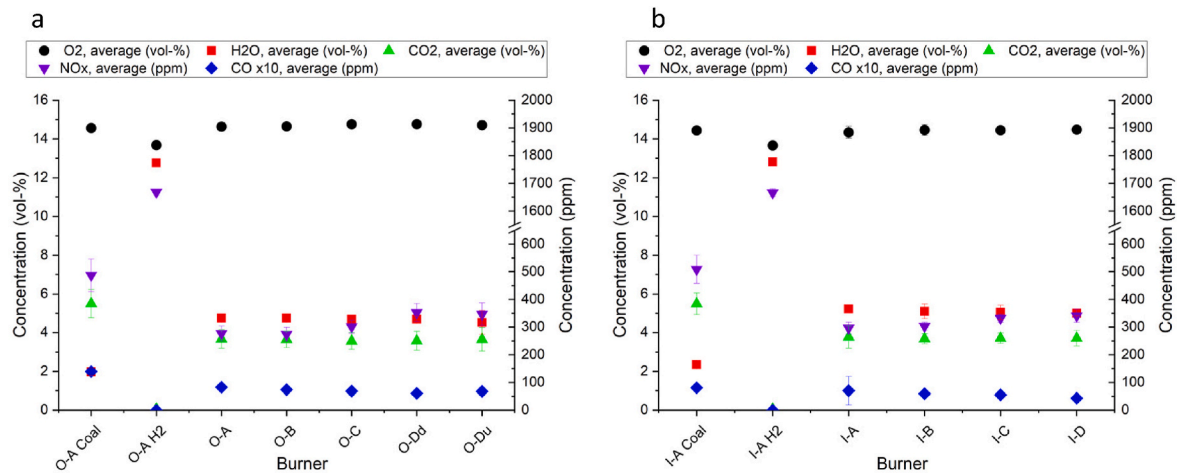


Fig. 9. Flue gas composition at the chimney using FTIR and MGA. (a) Corresponds to burner O measurements. (b) Corresponds to burner I measurements.

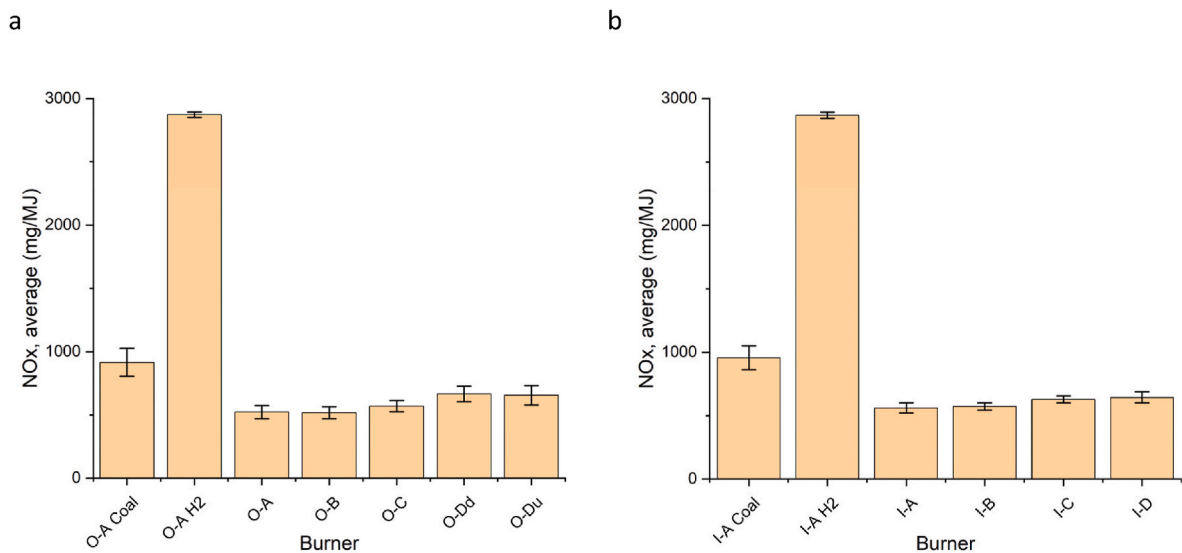


Fig. 10. Average mass of NO_x emitted at the chimney per unit of energy (a) for burner O (b) for burner I.

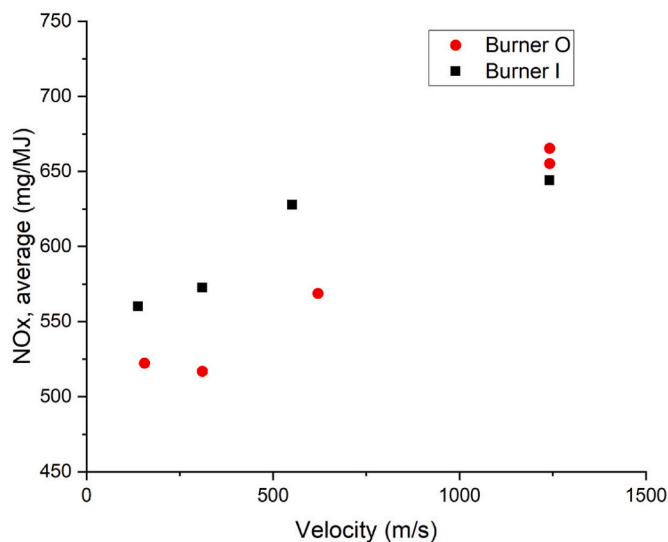


Fig. 11. Average NO_x emissions of the co-fired burners as a function of H₂ injection velocity.

editing, Writing – original draft, Visualization, Validation, Methodology, Investigation, Formal analysis, Data curation, Conceptualization. **Alexey Sepman:** Writing – review & editing, Visualization, Validation, Methodology, Formal analysis, Data curation. **Samuel Colin:** Writing – review & editing. **Jonas Wennebro:** Writing – review & editing, Project supervision. **Fredrik Normann:** Writing – review & editing, Project administration, Funding acquisition. **Henrik Wiinikka:** Writing – review & editing, Supervision, Project administration, Methodology, Funding acquisition, Conceptualization.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.ijhydene.2024.09.327>.

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